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                 Zentralblatt
NEWS 3 OCT 19 BEILSTEIN updated with new compounds
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NEWS 5 NOV 19 WPIX enhanced with XML display format
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                 MEDLINE segment
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NEWS 15 DEC 17 STN Viewer enhanced with full-text patent content
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NEWS 16 JAN 02
                 STN pricing information for 2008 now available
NEWS 17 JAN 16 CAS patent coverage enhanced to include exemplified
                 prophetic substances
NEWS 18 JAN 28 USPATFULL, USPAT2, and USPATOLD enhanced with new
                 custom IPC display formats
NEWS 19 JAN 28 MARPAT searching enhanced
NEWS 20 JAN 28 USGENE now provides USPTO sequence data within 3 days
                 of publication
NEWS 21 JAN 28 TOXCENTER enhanced with reloaded MEDLINE segment
NEWS 22 JAN 28 MEDLINE and LMEDLINE reloaded with enhancements
NEWS 23 FEB 08 STN Express, Version 8.3, now available
NEWS 24 FEB 20 PCI now available as a replacement to DPCI
NEWS 25 FEB 25 IFIREF reloaded with enhancements
NEWS 26 FEB 25 IMSPRODUCT reloaded with enhancements
NEWS 27 FEB 29 WPINDEX/WPIDS/WPIX enhanced with ECLA and current
                 U.S. National Patent Classification
NEWS EXPRESS FEBRUARY 08 CURRENT WINDOWS VERSION IS V8.3,
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AND CURRENT DISCOVER FILE IS DATED 20 FEBRUARY 2008

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FULL ESTIMATED COST

SINCE FILE TOTAL
ENTRY SESSION
2.52 2.52

FILE 'REGISTRY' ENTERED AT 07:34:29 ON 14 MAR 2008
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STRUCTURE FILE UPDATES: 13 MAR 2008 HIGHEST RN 1007849-81-1 DICTIONARY FILE UPDATES: 13 MAR 2008 HIGHEST RN 1007849-81-1

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TSCA INFORMATION NOW CURRENT THROUGH January 9, 2008.

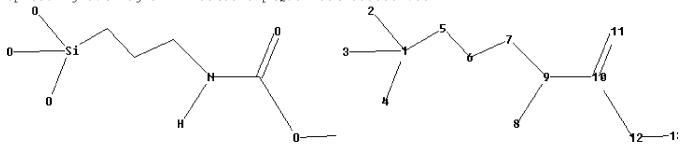
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http://www.cas.org/support/stngen/stndoc/properties.html

=>

Uploading C:\Program Files\Stnexp\Queries\155ester.str



chain nodes :

1 2 3 4 5 6 7 8 9 10 11 12 13

chain bonds :

1-2 1-3 1-4 1-5 5-6 6-7 7-9 8-9 9-10 10-11 10-12 12-13

exact/norm bonds :

7-9 9-10 10-11 10-12 12-13

exact bonds :

1-2 1-3 1-4 1-5 5-6 6-7 8-9

Match level :

1:CLASS 2:CLASS 3:CLASS 4:CLASS 5:CLASS 6:CLASS 7:CLASS 8:CLASS 9:CLASS 10:CLASS 11:CLASS 12:CLASS 13:CLASS

L1 STRUCTURE UPLOADED

=> d 11

L1 HAS NO ANSWERS

L1 STR

Structure attributes must be viewed using STN Express query preparation.

=> s 11

SAMPLE SEARCH INITIATED 07:34:58 FILE 'REGISTRY'
SAMPLE SCREEN SEARCH COMPLETED - 115 TO ITERATE

100.0% PROCESSED 115 ITERATIONS

INCOMPLETE SEARCH (SYSTEM LIMIT EXCEEDED)

SEARCH TIME: 00.00.01

FULL FILE PROJECTIONS: ONLINE **COMPLETE**

BATCH **COMPLETE**

PROJECTED ITERATIONS: 1657 TO 2943 PROJECTED ANSWERS: 640 TO 1520

L2 50 SEA SSS SAM L1

=>

Uploading C:\Program Files\Stnexp\Queries\155amide.str

$$\begin{array}{c} 0 \\ 0 \\ 0 \\ 0 \\ \end{array}$$

50 ANSWERS

chain nodes :

1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16

chain bonds :

 $1-2 \quad 1-3 \quad 1-4 \quad 1-5 \quad 5-6 \quad 6-7 \quad 7-8 \quad 8-9 \quad 8-15 \quad 10-12 \quad 10-11 \quad 10-16 \quad 12-13 \quad 12-14$

15-16

exact/norm bonds :

7-8 8-15 10-12 10-11

exact bonds :

1-2 1-3 1-4 1-5 5-6 6-7 8-9 10-16 12-13 12-14 15-16

Match level:

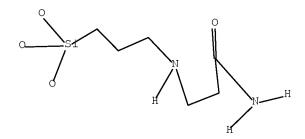
1:CLASS 2:CLASS 3:CLASS 4:CLASS 5:CLASS 6:CLASS 7:CLASS 8:CLASS 9:CLASS 10:CLASS 11:CLASS 12:CLASS 13:CLASS 14:CLASS 15:CLASS 16:CLASS

L3 STRUCTURE UPLOADED

=> d 13

L3 HAS NO ANSWERS

L3 STR



Structure attributes must be viewed using STN Express query preparation.

=> s 13

SAMPLE SEARCH INITIATED 07:35:36 FILE 'REGISTRY'

SAMPLE SCREEN SEARCH COMPLETED - 11 TO ITERATE

100.0% PROCESSED 11 ITERATIONS 0 ANSWERS

SEARCH TIME: 00.00.01

FULL FILE PROJECTIONS: ONLINE **COMPLETE**

BATCH **COMPLETE**

PROJECTED ITERATIONS: 22 TO 418 PROJECTED ANSWERS: 0 TO 0

L4 0 SEA SSS SAM L3

=> s 13 full

FULL SEARCH INITIATED 07:35:52 FILE 'REGISTRY'

FULL SCREEN SEARCH COMPLETED - 256 TO ITERATE

100.0% PROCESSED 256 ITERATIONS

SEARCH TIME: 00.00.01

L5 5 SEA SSS FUL L3

=> FIL HCAPLUS USPATFULL BIOSIS MEDLINE

COST IN U.S. DOLLARS SINCE FILE TOTAL

FULL ESTIMATED COST ENTRY SESSION 179.28 181.80

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FILE 'MEDLINE' ENTERED AT 07:36:02 ON 14 MAR 2008

=> s 15

L6 7 L5

=> dup rem 16

PROCESSING COMPLETED FOR L6

L7 7 DUP REM L6 (0 DUPLICATES REMOVED)

=> d 17 1-7 ibib abs hitstr

L7 ANSWER 1 OF 7 HCAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: 2005:1319723 HCAPLUS <u>Full-text</u>

DOCUMENT NUMBER: 144:219440

TITLE: Preparation and evaluation of novel stationary phases

for improved chromatographic purification of

pneumocandin B0

AUTHOR(S): Welch, Christopher J.; DaSilva, Jimmy O.; Nti-Gyabaah,

Joseph; Antia, Firoz; Goklen, Kent; Boyd, Russell

CORPORATE SOURCE: Merck Research Laboratories, Merck & Co. Inc., Rahway,

NJ, 07065, USA

SOURCE: Journal of Chromatography, A (2006), 1101(1-2),

204-213

CODEN: JCRAEY; ISSN: 0021-9673

PUBLISHER: Elsevier B.V.

DOCUMENT TYPE: Journal LANGUAGE: English

AB Prepn. and evaluation of a no. of stationary phases for improved chromatog. purification of pneumocandin B0, a key intermediate in the synthesis of the antifungal agent, Cancidas, has led to the identification of several materials with potential for improved performance.

IT 18388-80-2

RL: AMX (Analytical matrix); ANST (Analytical study)

(preparation and evaluation of novel stationary phases for improved chromatog. purification of pneumocandin B0)

RN 18388-80-2 HCAPLUS

CN Propanamide, 3-[[3-(triethoxysilyl)propyl]amino]- (CA INDEX NAME)

$$H_2N$$
— C — CH_2 — CH_2 — NH — $(CH_2)_3$ — S i— OEt

REFERENCE COUNT: 17 THERE ARE 17 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L7 ANSWER 2 OF 7 USPATFULL on STN

ACCESSION NUMBER: 2003:332278 USPATFULL Full-text

TITLE: Inorganic-organic hybrid gels for extracting species

such as lanthanides and actinides, and their

preparation

INVENTOR(S): Meyer, Daniel, St Genier de Comolas, FRANCE

Conocar, OLivier, Angles, FRANCE Moreau, Joel, Montpellier, FRANCE

Wong Chi Man, Michel, Vendargues, FRANCE

PATENT ASSIGNEE(S): Commissariat a l'Energie Atomique and Compagnie

Generale des Matieres Nucleaires, Velizy Villascoublay,

FRANCE (non-U.S. corporation)

	NUMBER	KIND	DATE	
-				
PATENT INFORMATION: U	JS 6667016	В1	20031223	
V	√O 9921654		19990506	
APPLICATION INFO.: U	JS 2000-530133		20001006	(9)
V	NO 1998-FR9902309		19981028	

NUMBER	DATE
ED 1007 12565	10071020

PRIORITY INFORMATION: FR 1997-13565 19971029

DOCUMENT TYPE: Utility
FILE SEGMENT: GRANTED
PRIMARY EXAMINER: Bos, Steven

LEGAL REPRESENTATIVE: Burns, Doane, Swecker & Mathis, L.L.P.

NUMBER OF CLAIMS: 10 EXEMPLARY CLAIM: 1

NUMBER OF DRAWINGS: 4 Drawing Figure(s); 2 Drawing Page(s)

LINE COUNT: 1111

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB Inorganic-organic hybrid gels can be employed to extract chemical species

such as lanthanides and actinides and their preparation.

The gels include a network of inorganic components with the formula: ##STR1##

where M is Si, Ti, Zr or Al, wherein organic molecules which can complex the species to be extracted are integrated, with each organic molecule being covalently linked to one or several M atoms in the network. The network is made by a sol-gel process from metallic alkoxides functionalized with complexing groups (amino, ether, hydroxy, amido, pyridino and bipyridino), capable of extracting metals such as lanthanides and actinides.

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

IT 223476-23-1DP, hydrolysis-condensation products with

tetraethoxysilane

(inorg.-organic functionalized hybrid gels for extraction of lanthanides

and

actinides)

RN 223476-23-1 USPATFULL

CN Propanediamide, N-[3-(triethoxysilyl)propyl]- (9CI) (CA INDEX NAME)

IT 223476-23-1

(inorg.-organic functionalized hybrid gels for extraction of lanthanides

and

actinides)

RN 223476-23-1 USPATFULL

CN Propanediamide, N-[3-(triethoxysilyl)propyl]- (9CI) (CA INDEX NAME)

L7 ANSWER 3 OF 7 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2001:354297 HCAPLUS Full-text

DOCUMENT NUMBER: 135:108228

TITLE: Hybrid silica based materials as new solid phase

extractants

AUTHOR(S): Bourg, Stephane; Broudic, Jean-Charles; Conocar,

Olivier; Moreau, Joel J. E.; Meyer, Daniel; Man,

Michel Wong Chi

CORPORATE SOURCE: DCC/DRRV/SPHA, CEA Valrho, DCC/DRRV/SPHA, Bagnols sur

Ceze, 30207, Fr.

SOURCE: Materials Research Society Symposium Proceedings

(2001), 628 (Organic/Inorganic Hybrid Materials),

CC1.6.1-CC1.6.11

CODEN: MRSPDH; ISSN: 0272-9172

PUBLISHER: Materials Research Society

DOCUMENT TYPE: Journal LANGUAGE: English

Org.-inorg. hybrids exhibiting specific properties are easily prepd. by incorporation of organic fragments in an inorg. network. Increasing attention is being paid to hybrid silsesquioxane gels, which are prepared by sol-gel hydrolysis condensation of organic mols. containing two or more trialkoxysilyl substituents. These hybrids consist of a mixed three-dimensional network, where the organic fragment, crosslinking siloxane chains, is part of the framework. Owing to the presence of a strong Si-C bond between the organic and the inorg. fragments, highly stable hybrid network are produced in this way. A variety of materials can be produced according to the intrinsic properties of the organic We report here the preparation of hybrid materials with complexing properties upon hydrolysis-condensation of ligands

functionalized by Si(OR)3 groups. New hybrid silica based materials containing malonamide ligands have been prepared by sol-gel hydrolysis condensation of functionalized precursors and have been used as solid phase extractants for the complexation of actinides. This approach is quite different from the classical immobilization procedure of complexing agents. The sol-gel approach allows one to adjust the ligand loading and to achieve some control and some tuning of the ligand environment since the oxide matrix is built around the complexing moieties. These hybrids proved to be highly efficient extracting solids.

IT 349649-16-7P

RL: NUU (Other use, unclassified); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)

(extractants; hybrid silica based materials as new solid phase
extractants)

RN 349649-16-7 HCAPLUS

CN Silicic acid (H4SiO4), tetraethyl ester, polymer with N-[3- (triethoxysilyl)propyl]propanediamide (9CI) (CA INDEX NAME)

CM 1

CRN 223476-23-1 CMF C12 H26 N2 O5 Si

CM 2

CRN 78-10-4 CMF C8 H20 O4 Si

REFERENCE COUNT: 24 THERE ARE 24 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L7 ANSWER 4 OF 7 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1999:297351 HCAPLUS Full-text

DOCUMENT NUMBER: 130:298824

TITLE: Inorganic-organic hybrid gels for extracting species

such as lanthanides and actinides from aqueous

solutions, and their preparation

INVENTOR(S): Meyer, Daniel; Conocar, Olivier; Moreau, Joel; Wong

Chi Man, Michel

PATENT ASSIGNEE(S): Commissariat a l'Energie Atomique, Fr.; Compagnie

Generale des Matieres Nucleaires

SOURCE: PCT Int. Appl., 86 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: French

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PAT	ENT NO.			KINI)	DATE	API	PLICATION NO.		DATE
WO	9921654		T/D	A1		19990506	WO	1998-FR2309		19981028
₽D.	W: GE 2770153	B, JP,	KK,	RU, A1	UA,	, US 19990430	гD	1997-13565		19971029
	2770153			B1		19991119	L IX	1997-13303		133/1023
	2349347			A		20001101	GB	2000-9736		19981028
GB .	2349347	7		В		20011114				
JP .	2001520	932		${f T}$		20011106	JΡ	2000-517801		19981028
RU .	2207190)		C2		20030627	RU	2000-113194		19981028
US	6667016)		В1		20031223	US	2000-530133		20001006
PRIORITY	APPLN.	INFO	.:				FR	1997-13565	A	19971029
							WO	1998-FR2309	W	19981028
							WO	1999-FR2309	W	19981028

AB Inorg.-org. hybrid gels are described for extg. species such as lanthanides and actinides from aqueous solns. The gels comprise a system of inorg. units of formula =M-O- in which M is Si, Ti, Zr or Al, which are integrated mols. complexing the species to be extracted, each organic mol. being covalently bound to one or several M atoms of the system. The gels are prepared by solgel processing (i.e. hydrolysis and catalytic polycondensation) of metal alkoxides functionalized with complexing groups (amino, ether, hydroxy, amido, pyridino and bipyridino), capable of extracting metals such as lanthanides and actinides. In examples, an NH4F catalyst in EtOH/H2O was used for hydrolysis and polycondensation of functionalized silicon alkoxides with tetraethoxysilane. One silicon alkoxide was 4-[2-(triethoxysilyl)ethyl]pyridine.

IT 223476-23-1DP, hydrolysis-condensation products with

tetraethoxysilane

RL: NUU (Other use, unclassified); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)

(inorg.-organic functionalized hybrid gels for extraction of lanthanides

and

actinides)

RN 223476-23-1 HCAPLUS

CN Propanediamide, N-[3-(triethoxysilyl)propyl]- (9CI) (CA INDEX NAME)

IT 223476-23-1

RL: RCT (Reactant); RACT (Reactant or reagent)

(inorg.-organic functionalized hybrid gels for extraction of lanthanides

and

actinides)

RN 223476-23-1 HCAPLUS

CN Propanediamide, N-[3-(triethoxysilyl)propyl]- (9CI) (CA INDEX NAME)

$$H_2N$$
— $\overset{\circ}{U}$ — CH_2 — $\overset{\circ}{U}$ — NH — $(CH_2)_3$ — $\overset{\circ}{S}$ i— \circ Et

REFERENCE COUNT: 12 THERE ARE 12 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L7 ANSWER 5 OF 7 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1999:645493 HCAPLUS Full-text

DOCUMENT NUMBER: 132:36584

TITLE: New hybrid silica based materials for the solid-liquid

extraction of actinides

AUTHOR(S): Broudic, Jean-Charles; Conocar, Olivier; Moreau, Joel

J. E.; Meyer, Daniel; Wong Chi Man, Michel

CORPORATE SOURCE: DCC/DRRV/SPHA, CEA-Valrho, Bagnols sur Ceze, 30207,

Fr.

SOURCE: Journal of Materials Chemistry (1999), 9(10),

2283-2285

CODEN: JMACEP; ISSN: 0959-9428 Royal Society of Chemistry

DOCUMENT TYPE: Journal LANGUAGE: English

PUBLISHER:

AB New hybrid silica-based materials contg. malonamide ligands were prepd. by sol-gel hydrolysis condensation of functionalized precursors. Extraction properties of solid hybrids were tested for aqueous nitric acid solns. of Pu(IV) and Am (III). These hybrids proved to be highly efficient extracting solids which compare favorably to related silica materials.

IT 252337-31-8P 252337-34-1P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(hybrid silica based materials for the solid-liquid extraction of actinides) $\ensuremath{\mathsf{S}}$

RN 252337-31-8 HCAPLUS

CN Poly[[1,3-bis[3-[(3-amino-1,3-dioxopropyl)amino]propyl]-1,3:1,3-disiloxanediylidene]-1,3-bis(oxy)] (9CI) (CA INDEX NAME)

PAGE 1-A

H2N-C-CH2-C-NH-(GH2)3-Si-O-Si-(CH2)3-NH-C-CH2-C-

PAGE 1-B

RN 252337-34-1 HCAPLUS

CN Propanediamide, N-[3-(triethoxysily1)propy1]-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 223476-23-1 CMF C12 H26 N2 O5 Si

IT 223476-23-1P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (hybrid silica-based materials for the solid-liquid extraction of actinides)

RN 223476-23-1 HCAPLUS

CN Propanediamide, N-[3-(triethoxysilyl)propyl]- (9CI) (CA INDEX NAME)

REFERENCE COUNT: 23 THERE ARE 23 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L7 ANSWER 6 OF 7 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1962:469378 HCAPLUS Full-text

DOCUMENT NUMBER: 57:69378
ORIGINAL REFERENCE NO.: 57:13804f-i

TITLE: Aminoalkylsilicon compounds

INVENTOR(S): Pike, Ronald M.; Morehouse, Edward L.

PATENT ASSIGNEE(S): Union Carbide Corp.

SOURCE: 11 pp.

DOCUMENT TYPE: Patent
LANGUAGE: Unavailable

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 3033815		19620508	US 1959-836623	19590828
PRIORITY APPLN. INFO	0.:		US	19590828

AB Aminoalkvlsilicon compds. of the formula (I) H2NCaH2a-SiRbY3-b (a = at least 3, b = 0 to 2, Y = alkoxy radicals, R = alkyl or aryl) react with BHC: CR'X(II) (B = hydrogen, alkyl or aryl; R' = hydrogen or alkyl; X = nitrile or COD, wherein D is hydrogen, alkyl, aryl, alkoxy, or amino) forming organosilicon compds. having at least one group of the formula XCHR'CHBNHCaH2aSiRbY3-b and (XCHR'CHB)2NCaH2aSiRbY3-b. To 75 g. γ -aminopropyltriethoxysilane was added 29.2 g. Me acrylate and the mixture was

heated under constant stirring at $80^{\circ}/2$ mm. Distillation of the product 61.4q. γ-(N-2-carbomethoxyethyl)aminopropyltriethoxysilane (III), b0..33-0.38 109-11°, n25D 1.4308, and 3.6 g. γ -N,N- bis(2 -

carbomethoxyethyl)aminopropyltriethoxysilane, b0.0,-0.9 145-67°, n25D 1.4388, besides two unidentified fractions b0.55-0.38 $55-104^{\circ}$ and b0.33-0.30 $130-140^{\circ}$. Hydrolysis of 32 g. III with 18 g. H2O and 15 ml. concentrated HCl at 33°, with a stream of argon passing through the solution, yielded after evaporation of the liquid products (2 hrs. at $100^{\circ}/1-5$ mm.) 24.7 g. γ , -(N-2carbomethoxyethyl)aminopropylpolysiloxane, a white, resinous product. I may also be an aminoalkylsiloxane polymer or a siloxane copolymer containing the unit H2NCaH2aSiRbO(3-b)/2, forming with II (XCHR'CHB)2NCaH2aSiRbO(3-b)/2 and XCHRtCHBNHCaH2aSiRbO(3-b/2. The products are useful as sizes for fibrous materials, particularly fibrous glass materials, and as adhesives and flocculation agents.

18388-80-2P, Propionamide, 3-[[3-(triethoxysilyl)propyl]amino]-ΤT RL: PREP (Preparation)

(preparation of)

- RN 18388-80-2 HCAPLUS
- Propanamide, 3-[[3-(triethoxysilyl)propyl]amino]- (CA INDEX NAME) CN

ANSWER 7 OF 7 HCAPLUS COPYRIGHT 2008 ACS on STN L7

ACCESSION NUMBER: 1962:60684 HCAPLUS

DOCUMENT NUMBER: 56:60684 ORIGINAL REFERENCE NO.: 56:11621d-q

TITLE: Organosilicon compounds and process for producing same

INVENTOR(S): Pike, Ronald Marston; Morehouse, Edward L.

PATENT ASSIGNEE(S): Union Carbide Corp.

DOCUMENT TYPE: Pat.ent. LANGUAGE: Unavailable

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
GB 882051		19611108	GB 1957-30940	19571003
DE 1114326			DE	
PRIORITY APPLN. INFO.:			US	19561012
			GB	19571003

Organosilicon compds. contq. substituted amino groups linked to the Si atom AΒ through a polymethylene linkage of at least 3 C atoms are prepared by the reaction of an aminoalkyl Si compound with an α, β - olefinically unsatd. organic compound at $80-80^{\circ}$. (γ - Aminopropyl)triethoxy-silane (I) (75 g.) and 29.2 g. Me acrylate were stirred together at 2 mm. to a temperature of 80°. The product was fractionally distilled The fraction b0.33-0.38 109-11° was 61.4 g. $[\gamma-(N-2-carbomethoxyethyl)]$ aminopropyl]triethoxysilane, n25 1.4308. Prepared similarly was $[\gamma-(N,N-di-2-carbethoxy$ ethyl)aminopropyl]triethoxysilane, b0.4-0.45 149-66°, n2D5 1.4372-1.4379. Acrylamide (89.1 g.) is added dropwise to 110.7 g. I with stirring, the mixture heated to 80° 4 hrs., distilled in vacuo until 49.5 g. was collected. The fraction b1.52-2.5 85-192°, n25 1.4448-1.4521 was [γ -(N-2-aminoethylaminopropyl]triethoxysilane. I (442.6 g.) under argon was cooled to 5°, 213.4 g. acrylonitrile added dropwise below 30°, the mixture kept overnight, and a 327.4-g. portion distilled; the 210.6-g. fraction b0.6-0.7 127-32° was [γ -(N-2-cyanoethyl)-aminopropyl]triethoxysilane. Prepared similarly were: [δ -(N-2-cyanoethyl)aminobutyl]triethoxysilane, b0.3-0.4 128-35°, n2D5 1 4370; [δ -(N-2-cyanoethyl)butyl]methyldiethoxy silane, b0.9 115-16°, n2D5 1.4423; [δ -(N-1-phenyl-2-carbethoxyethyl)-aminobutyl]methyldiethoxysilane, b0.5152-62°, n25 1.4776.

IT 18388-80-2P, Propionamide, 3-[[3-(triethoxysilyl)propyl]amino]-RL: PREP (Preparation)

(preparation of)

RN 18388-80-2 HCAPLUS

CN Propanamide, 3-[[3-(triethoxysilyl)propyl]amino]- (CA INDEX NAME)

=> s 17 and silica

L8 5 L7 AND SILICA

=> d 18 ibib abs

L8 ANSWER 1 OF 5 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2005:1319723 HCAPLUS Full-text

DOCUMENT NUMBER: 144:219440

TITLE: Preparation and evaluation of novel stationary phases

for improved chromatographic purification of

pneumocandin B0

AUTHOR(S): Welch, Christopher J.; DaSilva, Jimmy O.; Nti-Gyabaah,

Joseph; Antia, Firoz; Goklen, Kent; Boyd, Russell

CORPORATE SOURCE: Merck Research Laboratories, Merck & Co. Inc., Rahway,

NJ, 07065, USA

SOURCE: Journal of Chromatography, A (2006), 1101(1-2),

204-213

CODEN: JCRAEY; ISSN: 0021-9673

PUBLISHER: Elsevier B.V.

DOCUMENT TYPE: Journal LANGUAGE: English

AB Prepn. and evaluation of a no. of stationary phases for improved chromatog. purification of pneumocandin B0, a key intermediate in the synthesis of the antifungal agent, Cancidas, has led to the identification of several materials with potential for improved performance.

REFERENCE COUNT: 17 THERE ARE 17 CITED REFERENCES AVAILABLE FOR THIS

RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

=> d 18 1-5 ibib abs

L8 ANSWER 1 OF 5 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2005:1319723 HCAPLUS Full-text

DOCUMENT NUMBER: 144:219440

TITLE: Preparation and evaluation of novel stationary phases

for improved chromatographic purification of

pneumocandin B0

AUTHOR(S): Welch, Christopher J.; DaSilva, Jimmy O.; Nti-Gyabaah,

Joseph; Antia, Firoz; Goklen, Kent; Boyd, Russell

CORPORATE SOURCE: Merck Research Laboratories, Merck & Co. Inc., Rahway,

NJ, 07065, USA

SOURCE: Journal of Chromatography, A (2006), 1101(1-2),

204-213

CODEN: JCRAEY; ISSN: 0021-9673

PUBLISHER: Elsevier B.V.

DOCUMENT TYPE: Journal LANGUAGE: English

AB Prepn. and evaluation of a no. of stationary phases for improved chromatog. purification of pneumocandin B0, a key intermediate in the synthesis of the antifungal agent, Cancidas, has led to the identification of several materials

with potential for improved performance.

REFERENCE COUNT: 17 THERE ARE 17 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L8 ANSWER 2 OF 5 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2001:354297 HCAPLUS Full-text

DOCUMENT NUMBER: 135:108228

TITLE: Hybrid silica based materials as new solid

phase extractants

AUTHOR(S): Bourg, Stephane; Broudic, Jean-Charles; Conocar,

Olivier; Moreau, Joel J. E.; Meyer, Daniel; Man,

Michel Wong Chi

CORPORATE SOURCE: DCC/DRRV/SPHA, CEA Valrho, DCC/DRRV/SPHA, Bagnols sur

Ceze, 30207, Fr.

SOURCE: Materials Research Society Symposium Proceedings

(2001), 628(Organic/Inorganic Hybrid Materials),

CC1.6.1-CC1.6.11

CODEN: MRSPDH; ISSN: 0272-9172

PUBLISHER: Materials Research Society

DOCUMENT TYPE: Journal LANGUAGE: English

AΒ Org.-inorg. hybrids exhibiting specific properties are easily prepd. by incorporation of organic fragments in an inorg. network. Increasing attention is being paid to hybrid silsesquioxane gels, which are prepared by sol-gel hydrolysis condensation of organic mols. containing two or more trialkoxysilyl substituents. These hybrids consist of a mixed three-dimensional network, where the organic fragment, crosslinking siloxane chains, is part of the framework. Owing to the presence of a strong Si-C bond between the organic and the inorg. fragments, highly stable hybrid network are produced in this way. A variety of materials can be produced according to the intrinsic properties of the organic We report here the preparation of hybrid materials with complexing properties upon hydrolysis-condensation of ligands functionalized by Si(OR)3 groups. New hybrid silica based materials containing malonamide ligands have been prepared by sol-gel hydrolysis condensation of functionalized precursors and have been used as solid phase extractants for the complexation of actinides. This approach is quite different from the classical immobilization procedure of complexing agents. The sol-gel approach allows one to adjust the ligand loading and to achieve some control and some tuning of the ligand environment since the oxide matrix is built around the complexing moieties. These hybrids proved to be highly efficient extracting solids.

REFERENCE COUNT: 24 THERE ARE 24 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L8 ANSWER 3 OF 5 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1999:645493 HCAPLUS Full-text

DOCUMENT NUMBER: 132:36584

TITLE: New hybrid silica based materials for the

solid-liquid extraction of actinides

AUTHOR(S): Broudic, Jean-Charles; Conocar, Olivier; Moreau, Joel

J. E.; Meyer, Daniel; Wong Chi Man, Michel

CORPORATE SOURCE: DCC/DRRV/SPHA, CEA-Valrho, Bagnols sur Ceze, 30207,

Fr.

SOURCE: Journal of Materials Chemistry (1999), 9(10),

2283-2285

CODEN: JMACEP; ISSN: 0959-9428 Royal Society of Chemistry

DOCUMENT TYPE: Journal LANGUAGE: English

PUBLISHER:

AB New hybrid silica-based materials contg. malonamide ligands were prepared by sol-gel hydrolysis condensation of functionalized precursors. Extraction properties of solid hybrids were tested for aqueous nitric acid solns. of Pu(IV) and Am (III). These hybrids proved to be highly efficient extracting solids which compare favorably to related silica materials.

REFERENCE COUNT: 23 THERE ARE 23 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L8 ANSWER 4 OF 5 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1999:297351 HCAPLUS Full-text

DOCUMENT NUMBER: 130:298824

TITLE: Inorganic-organic hybrid gels for extracting species

such as lanthanides and actinides from aqueous

solutions, and their preparation

INVENTOR(S): Meyer, Daniel; Conocar, Olivier; Moreau, Joel; Wong

Chi Man, Michel

PATENT ASSIGNEE(S): Commissariat a l'Energie Atomique, Fr.; Compagnie

Generale des Matieres Nucleaires

SOURCE: PCT Int. Appl., 86 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: French

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PA'	TENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO	9921654	A1	19990506	WO 1998-FR2309	19981028
	W: GB, JP, KR,	RU, UA,			
FR	2770153	A1	19990430	FR 1997-13565	19971029
FR	2770153	В1	19991119		
GB	2349347	A	20001101	GB 2000-9736	19981028
GB	2349347	В	20011114		
JP	2001520932	Τ	20011106	JP 2000-517801	19981028
RU	2207190	C2	20030627	RU 2000-113194	19981028
US	6667016	В1	20031223	US 2000-530133	20001006
PRIORIT	Y APPLN. INFO.:			FR 1997-13565 A	19971029
				WO 1998-FR2309 W	19981028
				WO 1999-FR2309 W	19981028

AB Inorg.-org. hybrid gels are described for extg. species such as lanthanides and actinides from aqueous solns. The gels comprise a system of inorg. units of formula =M-O- in which M is Si, Ti, Zr or Al, which are integrated mols. complexing the species to be extracted, each organic mol. being covalently bound to one or several M atoms of the system. The gels are prepared by solgel processing (i.e. hydrolysis and catalytic polycondensation) of metal

alkoxides functionalized with complexing groups (amino, ether, hydroxy, amido, pyridino and bipyridino), capable of extracting metals such as lanthanides and actinides. In examples, an NH4F catalyst in ${\rm EtOH/H2O}$ was used for hydrolysis and polycondensation of functionalized silicon alkoxides with

tetraethoxysilane. One silicon alkoxide was 4-[2-

(triethoxysilyl)ethyl]pyridine.

REFERENCE COUNT: 12 THERE ARE 12 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L8 ANSWER 5 OF 5 USPATFULL on STN

ACCESSION NUMBER: 2003:332278 USPATFULL Full-text

TITLE: Inorganic-organic hybrid gels for extracting species

such as lanthanides and actinides, and their

preparation

INVENTOR(S): Meyer, Daniel, St Genier de Comolas, FRANCE

Conocar, OLivier, Angles, FRANCE Moreau, Joel, Montpellier, FRANCE

Wong Chi Man, Michel, Vendargues, FRANCE

PATENT ASSIGNEE(S): Commissariat a l'Energie Atomique and Compagnie

Generale des Matieres Nucleaires, Velizy Villascoublay,

FRANCE (non-U.S. corporation)

		NUMBER	KIND	DATE	
PATENT INFORMATION:	US	6667016	B1	20031223	
	WO	9921654		19990506	
APPLICATION INFO.:	US	2000-530133		20001006	(9)
	WO	1998-FR9902309		19981028	

NUMBER DATE

PRIORITY INFORMATION: FR 1997-13565 19971029

DOCUMENT TYPE: Utility
FILE SEGMENT: GRANTED
PRIMARY EXAMINER: Bos, Steven

LEGAL REPRESENTATIVE: Burns, Doane, Swecker & Mathis, L.L.P.

NUMBER OF CLAIMS: 10 EXEMPLARY CLAIM: 1

NUMBER OF DRAWINGS: 4 Drawing Figure(s); 2 Drawing Page(s)

LINE COUNT: 1111

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB Inorganic-organic hybrid gels can be employed to extract chemical species such as lanthanides and actinides and their preparation.

The gels include a network of inorganic components with the formula: ##STR1##

where M is Si, Ti, Zr or Al, wherein organic molecules which can complex the species to be extracted are integrated, with each organic molecule being covalently linked to one or several M atoms in the network. The network is made by a sol-gel process from metallic alkoxides functionalized with complexing groups (amino, ether, hydroxy, amido, pyridino and bipyridino), capable of extracting metals such as lanthanides and actinides.

=> s 12 and silica

L9 24 L2 AND SILICA

=> dup rem 19

PROCESSING COMPLETED FOR L9

24 DUP REM L9 (0 DUPLICATES REMOVED)

 \Rightarrow d 110 1-24 ibib abs hitstr

L10 ANSWER 1 OF 24 HCAPLUS COPYRIGHT 2008 ACS on STN 2007:849797 HCAPLUS Full-text ACCESSION NUMBER:

DOCUMENT NUMBER: 147:486116

TITLE: Resolution of three important π -basic chiral

compounds on recently developed five π -acidic

chiral columns

AUTHOR(S): Ryoo, Jae Jeong; Park, Joo-Bo; Kwon, Hae Young; Lee,

Jung Whan

CORPORATE SOURCE: Department of Chemistry Education, Kyungpook National

University, Taequ, 702-701, S. Korea

Bulletin of the Korean Chemical Society (2007), 28(6), SOURCE:

1042-1044

CODEN: BKCSDE; ISSN: 0253-2964

PUBLISHER: Korean Chemical Society

DOCUMENT TYPE: Journal LANGUAGE: English

Three important chiral compds., 1,1'-bi-2-naphthol (BNO), 1,1'-binaphthyl-AΒ 2,2'-diamine (BNA), 2,2,2-trifluoro-1-(9-anthryl)ethanol (TFAE), were used for testing the five Pirkle-type π -acidic chiral stationary phases. The resolution results were compared to those of com. available chiral columns. In comparison of the enantiosepn. results of BNA and TFAE on the five CSPs and known com. columns, the chiral separation on a phenylglycine CSP showed the best resolution This phase could be useful in a large-scale separation of these three important chiral compds.

ΙT 445385-80-8D, silica-bound

RL: ARU (Analytical role, unclassified); PEP (Physical, engineering or chemical process); ANST (Analytical study); PROC (Process)

(resolution of π -basic chiral compds. on recently developed π -acidic chiral columns)

445385-80-8 HCAPLUS RN

Carbamic acid, N-[3-(triethoxysily1)propy1]-, (2R)-2-[(3,5-CN dinitrobenzoyl)amino]-2-phenylethyl ester (CA INDEX NAME)

Absolute stereochemistry.

REFERENCE COUNT: 21 THERE ARE 21 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L10 ANSWER 2 OF 24 HCAPLUS COPYRIGHT 2008 ACS on STN 2006:699661 HCAPLUS Full-text ACCESSION NUMBER:

DOCUMENT NUMBER: 145:211498 TITLE: Nonlinear optical polymer based on hydroxy

dye/inorganic composite and its preparation by sol-gel

method

INVENTOR(S): Xi, Hongxia; Li, Zhong; Xia, Qibin; Liang, Zhaoxi

PATENT ASSIGNEE(S): South China University of Technology, Peop. Rep. China

SOURCE: Faming Zhuanli Shenqing Gongkai Shuomingshu, 11 pp.

CODEN: CNXXEV

DOCUMENT TYPE: Patent LANGUAGE: Chinese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
CN 1800239	A	20060712	CN 2005-10101677	20051130
PRIORITY APPLN. INFO.:			CN 2005-10101677	20051130
GI				

$$\begin{bmatrix} D-B & & & \\ & & \\ & & \\ & & \end{bmatrix}$$

$$\begin{bmatrix} D-B & & \\ & & \\ & & \\ & & \\ & & \\ & & \end{bmatrix}$$

$$\begin{bmatrix} D-B & & \\ & &$$

AΒ The title nonlinear optical polymer based on hydroxy dye/inorg. composite has a structure I, wherein A is -CnH2n-, or -NR-CnH2n-; B is -CN, -NO2, -CF3, -SO2R, -R, -CH=CH-, -N=N-, -CH=N-N=CH-, or -CH=N-; when B is -CN, -NO2, -CF3, -SO2R, or -R, D is absent; and when B is -CH=CH-, -N=N-, -CH=N-N=CH-, or -CH=N-, D is -H, -CN, -NO2, -CF3, -SO2R, -R, substituted or unsubstituted aromatic ring, condensed aromatic ring, heterocycle, or condensed aromatic heterocycle; R is C1-C9 alkyl; n and m are integers of 1-9, and P is a natural number The preparation comprises mixing an organic solvent (such as n-hexane) 10-20 mL, a hydroxy-containing dye monomer 0.1 mol, an organosilane containing isocyanate (such as γ - isocyanatepropyltriethoxysilane) 0.1-0.15 mol, and triethylamine catalyst 0.3-0.5 mL, refluxing at 50-200 °C for 12-24 h in N2 atmosphere, and precipitating with anhydrous n-hexane to obtain solid powder of alkoxysilane dye containing hydroxy dye; mixing Et orthosilicate, prepared alkoxysilane dye, water, and ethanol at a molar ratio of 1:(0.8-1.5):(6-8):(4.5-5.5), adding dropwise 1 mol/L HNO3 under stirring to make the reactant mixture hydrolyze, sealing, stirring, and aging to obtain silics sol containing hydroxy chromophores. The polymer has good high temperature stability and easy film-forming, and can be used in the fields of phototelegraphy, photoelectronics, and optical information processing, etc.

IT 675129-70-1P

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)

(preparation of nonlinear optical polymer based on hydroxy dye/inorg. composite by sol-gel method)

RN 675129-70-1 HCAPLUS

CN Carbamic acid, [3-(trimethoxysily1)propy1]-, 2-[ethy1[4-[(4-nitropheny1)azo]pheny1]amino]ethyl ester (9CI) (CA INDEX NAME)

L10 ANSWER 3 OF 24 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2006:609442 HCAPLUS Full-text

DOCUMENT NUMBER: 146:219600

TITLE: Enantioseparation by sonochromatography

AUTHOR(S): Ryoo, Jae Jeong; Song, Young-Ae; Jeong, Young Han;

Hyun, Myung Ho; Park, Jung Hag; Lee, Wonjae

CORPORATE SOURCE: Dept. of Chemistry Education, Kyungpook National

University, Taegu, 702-701, S. Korea

SOURCE: Bulletin of the Korean Chemical Society (2006), 27(5),

637-641

CODEN: BKCSDE; ISSN: 0253-2964

PUBLISHER: Korean Chemical Society

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 146:219600

AB Although chiral sepn. was one of the main topics of chromatog. practice for over twenty-five years, it still presents many difficulties. The ultrasonic dependence of chiral resolution was studied at various temps. to improve resolution and reduce anal. time. The chiral resolution was performed on recently commercialized two HPLC chiral stationary phases (CSP 1 and CSP 2) with the analogs of racemic N-acylnaphthylethylamines and racemic amino acid derivs. as analytes. The CSP 1 was prepared from a (R)-N-(3,5-dinitrobenzoyl) phenylglycinol and the CSP 2 was prepared from a (S)-N-3,5-(dinitrobenzoyl) leucine. From the comparison of the chromatog. results under sonic condition with those under nonsonic condition, the ultrasound decreased the elution time in chiral chromatog. at all temps. and improved the enantioselectivity at high temperature (45, 50, 60°).

IT 445385-80-8D, reaction product with silica gel

RL: ARU (Analytical role, unclassified); NUU (Other use, unclassified); ANST (Analytical study); USES (Uses)

(chiral phase; HPLC enantiosepn. enhanced by ultrasound application)

RN 445385-80-8 HCAPLUS

CN Carbamic acid, N-[3-(triethoxysily1)propy1]-, (2R)-2-[(3,5-dinitrobenzoy1)amino]-2-phenylethyl ester (CA INDEX NAME)

Absolute stereochemistry.

REFERENCE COUNT: 28 THERE ARE 28 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L10 ANSWER 4 OF 24 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2005:547629 HCAPLUS Full-text

DOCUMENT NUMBER: 143:73181

TITLE: Site selectively tagged and templated molecularly

imprinted polymers for sensor applications

INVENTOR(S): Bright, Frank V.

PATENT ASSIGNEE(S): The Research Foundation of State University of New

York, USA

SOURCE: PCT Int. Appl., 34 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

P	ATENT	NO.			KIN	D	DATE			APPI	LICAT	ION :	NO.		D.	ATE	
M	200	 50566	 13		A2	_	2005	0623		WO 2	2004-	 US41	560		2	0041	208
	W:	ΑE,	AG,	AL,	AM,	ΑT,	ΑU,	ΑZ,	BA,	BB,	BG,	BR,	BW,	BY,	BZ,	CA,	CH,
		CN,	CO,	CR,	CU,	CZ,	DE,	DK,	DM,	DZ,	EC,	EE,	EG,	ES,	FI,	GB,	GD,
		GE,	GH,	GM,	HR,	HU,	ID,	IL,	IN,	IS,	JP,	KΕ,	KG,	KP,	KR,	KΖ,	LC,
		LK,	LR,	LS,	LT,	LU,	LV,	MA,	MD,	MG,	MK,	MN,	MW,	MX,	MZ,	NA,	NΙ,
		NO,	NΖ,	OM,	PG,	PH,	PL,	PT,	RO,	RU,	SC,	SD,	SE,	SG,	SK,	SL,	SY,
		ТJ,	TM,	TN,	TR,	TT,	TZ,	UA,	UG,	US,	UZ,	VC,	VN,	YU,	ZA,	ZM,	ZW
	RW	: BW,	GH,	GM,	KΕ,	LS,	MW,	MZ,	NA,	SD,	SL,	SZ,	TZ,	UG,	ZM,	ZW,	AM,
		ΑZ,	BY,	KG,	KΖ,	MD,	RU,	ТJ,	TM,	ΑT,	BE,	BG,	CH,	CY,	CZ,	DE,	DK,
		EE,	ES,	FΙ,	FR,	GB,	GR,	HU,	ΙE,	IS,	IT,	LT,	LU,	MC,	NL,	PL,	PT,
		RO,	SE,	SI,	SK,	TR,	BF,	ВJ,	CF,	CG,	CI,	CM,	GΑ,	GN,	GQ,	GW,	ML,
		MR,	NE,	SN,	TD,	ΤG											
Al	U 200	42972	74		A1		2005	0623		AU 2	2004-	2972	74		2	0041	208
C	A 254	7418			A1		2005	0623		CA 2	2004-	2547	418		2	0041	208
U	S 200	52272	58		A1		2005	1013		US 2	2004-	6857			2	0041	208
E	P 169	9832			A2		2006	0913		EP 2	2004-	8138	20		2	0041	208
	R:	ΑT,	BE,	CH,	DE,	DK,	ES,	FR,	GB,	GR,	IT,	LI,	LU,	NL,	SE,	MC,	PT,
		ΙE,	SI,	LT,	LV,	FΙ,	RO,	MK,	CY,	AL,	TR,	BG,	CZ,	EE,	HU,	PL,	SK,
		BA,	HR,	IS,	YU												
J1	P 200	75255	70		Τ		2007	0906		JP 2	2006-	5440	56		2	0041	208
PRIORI'	TY AP	PLN.	INFO	.:						US 2	2003-	5276	61P]	P 2	0031	208
										WO 2	2004-	US41	560	Ī	W 2	0041	208

AB This invention provides molecularly imprinted polymers (MIPs) for the detection of analytes, methods for forming the MIPs and detecting the analyte using the MIPs. The MIP comprises templated sites which are formed using a mimic of the analyte such that a reporter compound can be selectively attached at the templated sites, thus providing a site selectively tagged and templated MIP.

IT 855005-68-4P

RL: ARU (Analytical role, unclassified); IMF (Industrial manufacture); ANST (Analytical study); PREP (Preparation)

(xerogel; manufacture of site selectively tagged and templated molecularly imprinted polymers for sensor applications)

RN 855005-68-4 HCAPLUS

CN Carbamic acid, [3-(triethoxysily1)propy1]-, [[[[[3-(triethoxysily1)propy1]amino]carbony1]oxy]methylidyne]tri-4,1-phenylene ester, polymer with silicic acid (H4SiO4) tetraethyl ester, 2,2'-[[3-(triethoxysily1)propy1]imino]bis[ethano1] and trimethoxyoctylsilane (9CI) (CA INDEX NAME)

CM 1

CRN 855005-67-3

CMF C59 H100 N4 O20 Si4

PAGE 1-B

CM 2

CRN 7538-44-5 CMF C13 H31 N O5 Si

CM 3

CRN 3069-40-7 CMF C11 H26 O3 Si

CM 4

CRN 78-10-4 CMF C8 H20 O4 Si

L10 ANSWER 5 OF 24 USPATFULL on STN

ACCESSION NUMBER: 2005:261235 USPATFULL Full-text

TITLE: Site selectively tagged and templated molecularly

imprinted polymers for sensor applications

INVENTOR(S): Bright, Frank V., Williamsville, NY, UNITED STATES

APPLICATION INFO.: US 2004-6857 A1 20041208 (11)

NUMBER DATE

PRIORITY INFORMATION: US 2003-527661P 20031208 (60)

DOCUMENT TYPE: Utility
FILE SEGMENT: APPLICATION

LEGAL REPRESENTATIVE: HODGSON RUSS LLP, ONE M & T PLAZA, SUITE 2000, BUFFALO,

NY, 14203-2391, US

NUMBER OF CLAIMS: 32 EXEMPLARY CLAIM: 1

NUMBER OF DRAWINGS: 4 Drawing Page(s)

LINE COUNT: 1099

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

This invention provides molecularly imprinted polymers (MIPs) for the detection of analytes, methods for forming the MIPs and detecting the analyte using the MIPs. The MIP comprises templated sites which are formed using a mimic of the analyte such that a reporter compound can be selectively attached at the templated sites, thus providing a site selectively tagged and templated MIP.

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

IT 855005-68-4P

(xerogel; manufacture of site selectively tagged and templated molecularly imprinted polymers for sensor applications)

RN 855005-68-4 USPATFULL

CN Carbamic acid, [3-(triethoxysily1)propy1]-, [[[[[3-

(triethoxysily1)propy1]amino]carbony1]oxy]methylidyne]tri-4,1-phenylene
ester, polymer with silicic acid (H4SiO4) tetraethyl ester,
2,2'-[[3-(triethoxysily1)propy1]imino]bis[ethano1] and
trimethoxyoctylsilane (9CI) (CA INDEX NAME)

CM 1

CRN 855005-67-3 CMF C59 H100 N4 O20 Si4

PAGE 1-A

PAGE 1-B

CM 2

CRN 7538-44-5 CMF C13 H31 N O5 Si

CM 3

CRN 3069-40-7 CMF C11 H26 O3 Si

CM 4

CRN 78-10-4 CMF C8 H20 O4 Si

L10 ANSWER 6 OF 24 USPATFULL on STN

ACCESSION NUMBER: 2005:209747 USPATFULL Full-text

TITLE: Coating composition using a polyester resin composition

for forming a paint film on a substrate

INVENTOR(S): Ramesh, Swaminathan, Canton, MI, UNITED STATES

Green, Marvin L., Brighton, MI, UNITED STATES

	NUMBER	KIND	DATE	
PATENT INFORMATION:	US 2005182232	A1	20050818	
	US 7041777	В2	20060509	
APPLICATION INFO.:	US 2004-844629	A1	20040513	(10)

RELATED APPLN. INFO.: Continuation-in-part of Ser. No. US 2004-777698, filed

on 12 Feb 2004, PENDING

DOCUMENT TYPE: Utility
FILE SEGMENT: APPLICATION

LEGAL REPRESENTATIVE: BASF CORPORATION, ANNE GERRY SABOURIN, 26701 TELEGRAPH

ROAD, SOUTHFIELD, MI, 48034-2442, US

NUMBER OF CLAIMS: 41
EXEMPLARY CLAIM: 1
LINE COUNT: 1017

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB A coating composition for producing films having improved scratch and mar characteristics. The coating composition incorporates a polyester polycarbamate resin composition, a first cross-linking agent, and a cross-linkable resin to form the film. The resin composition is the reaction product of a first compound having a plurality of hydroxyl groups with a carbamate compound reactive with the hydroxyl groups of the first compound and added in an amount sufficient to form a carbamated intermediary. The carbamated intermediary has at least one primary carbamate group available for cross-linking and has unreacted hydroxyl groups. Then, a silyl compound having a terminal isocyanate group is reacted with the unreacted hydroxyl groups of the carbamated intermediary. The silyl compound also has silylalkoxy groups available for secondary cross-linking. The first cross-linking agent and the cross-linkable resin react with the primary carbamate groups and the silylalkoxy groups, respectively, to produce the film having improved scratch and mar characteristics.

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

IT 862375-32-4P

 $({\it scratch-resistant\ coating\ composition\ containing\ polyester\ polycarbamate\ component})$

RN 862375-32-4 USPATFULL

CN 1,2-Cyclohexanedicarboxylic acid, 1,1'-[2-[[[(2-

carboxycyclohexyl)carbonyl]oxy]methyl]-2-(8,8-dimethoxy-3-oxo-2,9-dioxa-4-aza-8-siladec-1-yl)-1,3-propanediyl] ester, triester with

1,2,3-propanetriol monocarbamate 1-neodecanoate (9CI) (CA INDEX NAME)

CM 1

CRN 862375-31-3

CMF C36 H57 N O17 Si

CM 2

CRN 79245-77-5

CMF C13 H26 O4

CCI IDS

CDES 8:ID, NEO

$$HO-CH_2-CH-CH_2-O-C-(C9H_19-neo)$$

CM 3

CRN 463-77-4

CMF C H3 N O2

IT 862375-32-4DP, reaction products with Resimene 747 and acrylic resin

 $(scratch-resistant\ coating\ composition\ containing\ polyester\ polycarbamate\ component)$

RN 862375-32-4 USPATFULL

CN 1,2-Cyclohexanedicarboxylic acid, 1,1'-[2-[[[(2-carboxycyclohexyl)carbonyl]oxy]methyl]-2-(8,8-dimethoxy-3-oxo-2,9-dioxa-4-aza-8-siladec-1-yl)-1,3-propanediyl] ester, triester with 1,2,3-propanetriol monocarbamate 1-neodecanoate (9CI) (CA INDEX NAME)

CM 1

CRN 862375-31-3 CMF C36 H57 N O17 Si

CM 2

CRN 79245-77-5 CMF C13 H26 O4 CCI IDS CDES 8:ID,NEO

CM 3

CRN 463-77-4 CMF C H3 N O2

ACCESSION NUMBER: 2004:796486 HCAPLUS Full-text

DOCUMENT NUMBER: 141:317191

TITLE: Silica sol composition, membrane electrode

assembly with proton-exchange membrane, and fuel cell

PATENT ASSIGNEE(S): Fuji Photo Film Co. Ltd., Japan

SOURCE: Eur. Pat. Appl., 50 pp.

CODEN: EPXXDW

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT	NO.		KINI	D D	DATE		I	APPLICATION NO.			DATE					
		•														
EP 1463	140		A2	2	20040)929	E	ΞP	2004-	7161			2	0040	325	
R:	AT, BE,	CH,	DE,	DK,	ES,	FR,	GB,	GR	, IT,	LI,	LU,	NL,	SE,	MC,	PT,	
	IE, SI,	LT,	LV,	FI,	RO,	MK,	CY,	AL	, TR,	BG,	CZ,	EE,	HU,	PL,	SK	
JP 2004	307814		A		20041	1104	·	JΡ	2003-	4326	63		2	0031	226	
US 2004	241522		A1	2	20041	1202	Ţ	JS	2004-	80768	39		2	0040	324	
PRIORITY APP	LN. INFO	·:					Ċ	JΡ	2003-	82369	9		A 2	0030	325	
							·	JΡ	2003-	82370	Э		A 2	0030	325	
							-	JΡ	2003-	4326	63		A 2	0031	226	

AB Provided are a proton-exchange membrane of which the ionic cond. is high and the methanol crossover is low, and a fuel cell of high power that comprises the proton-exchange membrane. The proton-exchange membrane has a structure of mesogen-containing organic mol. chains and a proton-donating group-containing group covalent-bonding to a silicon-oxygen three-dimensional crosslinked matrix, in which at least a part of the organic mol. chains are oriented to form an aggregate thereof; and the fuel cell comprises the membrane.

IT 765279-49-0P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

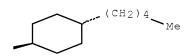
(silica sol composition, membrane electrode assembly with proton-exchange membrane, and fuel cell)

RN 765279-49-0 HCAPLUS

CN Carbamic acid, [3-(triethoxysilyl)propyl]-, 2-methyl-2-[[[8-[4-(trans-4-pentylcyclohexyl)phenoxy]octyl]oxy]methyl]-1,3-propanediyl ester (9CI) (CA INDEX NAME)

Relative stereochemistry.

PAGE 1-A



L10 ANSWER 8 OF 24 USPATFULL on STN

ACCESSION NUMBER: 2004:306853 USPATFULL Full-text

TITLE: Silica sol composition, membrane electrode

assembly with proton-exchange membrane, and fuel cell

INVENTOR(S): Ono, Michio, Minami-ashigara-shi, JAPAN

Wariishi, Koji, Minami-ashigara-shi, JAPAN Nomura, Kimiatsu, Minami-ashigara-shi, JAPAN Kikuchi, Wataru, Minami-ashigara-shi, JAPAN

PATENT ASSIGNEE(S): Fuji Photo Film Co., Ltd. (non-U.S. corporation)

DOCUMENT TYPE: Utility
FILE SEGMENT: APPLICATION

LEGAL REPRESENTATIVE: BRINKS HOFER GILSON & LIONE, P.O. BOX 10395, CHICAGO,

IL, 60610

NUMBER OF CLAIMS: 11
EXEMPLARY CLAIM: 1

NUMBER OF DRAWINGS: 3 Drawing Page(s)

LINE COUNT: 1989

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB Provided are a proton-exchange membrane of which the ionic conductivity is high and the methanol crossover is low, and a fuel cell of high power that comprises the proton-exchange membrane. The proton-exchange membrane has a structure of mesogen-containing organic molecular chains and a proton-donating group-containing group covalent-bonding to a silicon-oxygen three-dimensional crosslinked matrix, in which at least a part of the organic molecular chains are oriented to form an aggregate thereof; and the fuel cell comprises the membrane.

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

IT 765279-49-0P

(silica sol composition, membrane electrode assembly with proton-exchange membrane, and fuel cell)

RN 765279-49-0 USPATFULL

Relative stereochemistry.

PAGE 1-B

L10 ANSWER 9 OF 24 USPATFULL on STN

ACCESSION NUMBER: 2004:13532 USPATFULL Full-text

TITLE: Dental fillers, pastes, and compositions prepared

INVENTOR(S): Bui, Hoa T., Mendota Heights, MN, UNITED STATES

Kolb, Brant U., Afton, MN, UNITED STATES

Mitra, Sumita B., West St. Paul, MN, UNITED STATES

PATENT ASSIGNEE(S): 3M Innovative Properties Company, St. Paul, MN, UNITED

STATES, 55133-3427 (U.S. corporation)

	NUMBER	KIND	DATE	
PATENT INFORMATION:	US 2004010055	A1	20040115	
	US 7091259	В2	20060815	
APPLICATION INFO.:	US 2002-190321	A1	20020703	(10)
DOCUMENT TYPE:	Utility			
FILE SEGMENT:	APPLICATION			
LEGAL REPRESENTATIVE:	3M INNOVATIVE PRO		S COMPANY,	РО ВОХ
	PAUL, MN, 55133-3	3427		

X 33427, ST.

NUMBER OF CLAIMS: 41 EXEMPLARY CLAIM: 1 LINE COUNT: 1328

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

The present invention provides a dental paste, methods of making and using the dental paste, and compositions prepared therefrom. The dental paste includes a hardenable resin and an acid-reactive filler disposed in the resin. The acid-reactive filler includes a silane-treated surface, wherein the silane includes a poly(alkylene oxide) group-containing silane. Surfacetreated acid-reactive fillers, and methods of making and using the fillers, are also provided.

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

IT 643018-02-4P 643731-81-1P

(dental fillers and pastes containing silane-treated surfaces)

RN 643018-02-4 USPATFULL

CN Carbamic acid, [3-(triethoxysily1)propy1]-, 2-[2-(ethenyloxy)ethoxy]ethyl ester (9CI) (CA INDEX NAME)

RN 643731-81-1 USPATFULL

CN Poly[oxy(methyl-1,2-ethanediyl)], α -(2-methyl-1-oxo-2-propenyl)- ω -[[[[3-(triethoxysilyl)propyl]amino]carbonyl]oxy]- (9CI) (CA INDEX NAME)

$$\stackrel{\text{H2C}}{\text{Me}} \stackrel{\text{O}}{=} \stackrel{\text{O}}{\text{C}} \stackrel{\text{O}}{=} \stackrel{\text{O}}{\text{C}} \stackrel{\text{O}}{=} \stackrel{\text{O}}{\text{NH}} \stackrel{\text{O}}{=} \stackrel{\text{O}}{\text{CH2}} \stackrel{\text{O}}{\text{Si}} \stackrel{\text{O}}{=} \stackrel{\text{O}}{\text{Et}}$$

L10 ANSWER 10 OF 24 USPATFULL on STN

ACCESSION NUMBER: 2003:306367 USPATFULL Full-text

TITLE: Method for immobilizing oligonucleotides employing the

cycloaddition bioconjugation method

INVENTOR(S): Pieken, Wolfgang, Boulder, CO, UNITED STATES

Wolter, Andreas, Hamburg, GERMANY, FEDERAL REPUBLIC OF

Sebesta, David P., Longmont, CO, UNITED STATES Leuck, Michael, Boulder, CO, UNITED STATES

Latham-Timmons, Hallie A., Boulder, CO, UNITED STATES

Pilon, John, Ft. Collins, CO, UNITED STATES Husar, Gregory M., Longmont, CO, UNITED STATES

	NUMBER	KIND	DATE	
PATENT INFORMATION:	US 2003215801	A1	20031120	
APPLICATION INFO.:	US 2001-845742	A1	20010501	(9

RELATED APPLN. INFO.: Continuation-in-part of Ser. No. US 1999-341337, filed on 8 Jul 1999, PENDING A 371 of International Ser. No.

WO 1998-US649, filed on 8 Jan 1998, PENDING

		NUMB	ER	DATE	
PRIORITY	INFORMATION:	US 2000-201	561P	20000501	(60)
		US 2001-265	020P	20010130	(60)
DOCUMENT	TYPE .	IIt i l i t sz			

FILE SEGMENT: Utility
APPLICATION

LEGAL REPRESENTATIVE: SWANSON & BRATSCHUN L.L.C., 1745 SHEA CENTER DRIVE,

SUITE 330, HIGHLANDS RANCH, CO, 80129

NUMBER OF CLAIMS: 22 EXEMPLARY CLAIM: 1

NUMBER OF DRAWINGS: 12 Drawing Page(s)

LINE COUNT: 2144

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB This invention discloses a novel method for immobilizing molecules to a support. Specifically, this invention discloses a method of immobilizing derivatized biomolecules, such as oligonucleotides, using cycloaddition reactions, such as the Diels-Alder reaction. Included in this invention are the novel immobilized biomolecules that can be prepared according to the method of this invention.

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

IT 372108-00-4P

(method for immobilizing oligonucleotides employing the Diels Alder cycloaddn. bio-conjugation method)

RN 372108-00-4 USPATFULL

CN Carbamic acid, N-[3-(triethoxysily1)propy1]-, 9-anthracenylmethyl ester (CA INDEX NAME)

L10 ANSWER 11 OF 24 USPATFULL on STN

ACCESSION NUMBER: 2003:266159 USPATFULL Full-text

TITLE: Oxazolidine compound and curable resin composition

INVENTOR(S): Suga, Kazuo, Kanagawa, JAPAN

DOCUMENT TYPE: Utility FILE SEGMENT: APPLICATION

LEGAL REPRESENTATIVE: RADER FISHMAN & GRAUER PLLC, LION BUILDING, 1233 20TH

STREET N.W., SUITE 501, WASHINGTON, DC, 20036

NUMBER OF CLAIMS: 2
EXEMPLARY CLAIM: 1
LINE COUNT: 790

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB Disclosed is an oxazolidine compound represented by the following formula (1). ##STR1##

[R.sup.1 represents a hydrocarbon group having 1 to 6 carbon atoms; R.sup.2 and R.sup.3 independently represent a hydrogen atom or a monovalent hydrocarbon group having 1 to 15 carbon atoms, or R.sup.2 and R.sup.3 combine to form an alicyclic ring or an aromatic ring; R.sup.4 and R.sup.5 independently represent a methyl group or an ethyl group; R.sup.6 represents a single bond or --(R.sup.7--NHCOO--R.sup.8-- OCONH).sub.p--, R.sup.7 and R.sup.8 independently represent a divalent organic group, and p represents an integer of 1 to 15; m represents 2 or 3; n represents an integer of 1 to

3]. A curable resin composition using the oxazolidine compound as a latent curing agent have excellent curability and excellent storage stability, in particular, excellent storage stability in a state where the resin composition contains only the curable ingredient(s) and the latent curing agent.

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

IT 602325-94-0P

(silyl oxazolidines as latent hardeners for curable resin compns.)

RN 602325-94-0 USPATFULL

CN Carbamic acid, [3-(trimethoxysilyl)propyl]-, [2,3-dimethyl-2-(1-methylethyl)-5-oxazolidinyl]methyl ester, polymer with

2,2'-[(1-methylethylidene)bis(4,1-phenyleneoxymethylene)]bis[oxirane]
(9CI) (CA INDEX NAME)

CM 1

CRN 602325-92-8 CMF C16 H34 N2 O6 Si

CM 2

CRN 1675-54-3 CMF C21 H24 O4

$$\overset{\circ}{\longleftarrow} \mathtt{CH}_2 - \circ - \overset{\mathrm{Me}}{\longleftarrow} \mathtt{O} - \mathtt{CH}_2 \overset{\circ}{\longleftarrow}$$

L10 ANSWER 12 OF 24 USPATFULL on STN

ACCESSION NUMBER: 2003:74442 USPATFULL Full-text

TITLE: Powder coating or adhesives employing silanes or silane

treated fillers

INVENTOR(S): Katz, Lawrence E., Orange, CT, United States

Petty, Herbert E., Bethel, CT, United States

Su, Shiu-Chin Huang, Croton-on-Hudson, NY, United

States

Waldman, Bruce A., Peekskill, NY, United States Barbera, Bruce C., Beacon, NY, United States

PATENT ASSIGNEE(S): Crompton Corporation, Middlebury, CT, United States

(U.S. corporation)

	NUMBER	KIND	DATE	
PATENT INFORMATION: US	6534568	В1	20030318	
WO	9955754		19991104	
APPLICATION INFO.: US	2000-446609		20000419	(9)
WO	1999-US8694		19990421	

RELATED APPLN. INFO.: Continuation-in-part of Ser. No. US 1998-65911, filed on 24 Apr 1998, now abandoned Continuation-in-part of Ser. No. US 1998-115791, filed on 15 Jul 1998, now abandoned Continuation-in-part of Ser. No. US

1998-157658, filed on 21 Sep 1998, now abandoned

DOCUMENT TYPE: Utility FILE SEGMENT: GRANTED

PRIMARY EXAMINER: Dawson, Robert

ASSISTANT EXAMINER: Robertson, Jeffrey B. LEGAL REPRESENTATIVE: Dilworth, Michael P.

NUMBER OF CLAIMS: EXEMPLARY CLAIM:

NUMBER OF DRAWINGS: 0 Drawing Figure(s); 0 Drawing Page(s)

LINE COUNT: 1158

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

Powder coating or adhesive formulations include as a component thereof a AB silane of formula (I) or hydrolyzates or condensates thereof ##STR1##

where R.sup.1 is a hydrocarbon, acyl, alkylsilyl, or alkoxysilyl group, R.sup.2 is a monovalent hydrocarbon group; R.sup.3 is alkylene, optionally interrupted with one or more ether oxygen atoms; a is 0 or 1; Z is a direct bond or a divalent linking group; X is an m-valent organic group or H; and m is 1-20. The silanes are useful as crosslinkers, property modifiers and/or adhesion promoters. Powder adhesives may be similarly formed with the silanes of formula (I). Fillers or pigments, such as titanium dioxide fillers, which are treated with silanes, especially silanes having alkyl, epoxy, acryl, methacryl, polyether, amino, acid anhydride, hydroxyalkyl, carbamato or ureido functionality, may also be usefully employed in powder coating formulations.

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

IT 501925-94-6P 501925-95-7P

(carbamate silane-treated fillers)

RN 501925-94-6 USPATFULL

CN Carbamic acid, [3-(trimethoxysilyl)propyl]-, 1-ethyl-1,2-ethanediyl ester (9CI) (CA INDEX NAME)

501925-95-7 USPATFULL RN

Carbamic acid, [3-(trimethoxysily1)propy1]-, 1,3-propanediy1 ester (9CI) CN (CA INDEX NAME)

MeO—
$$\sin$$
 (CH₂)₃—NH— \cos (CH₂)₃—O— \cos NH—(CH₂)₃— \sin OMe ome

L10 ANSWER 13 OF 24 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2003:476005 HCAPLUS Full-text

DOCUMENT NUMBER: 139:277384

TITLE: New fully polymeric proton solvents with high proton

mobility

AUTHOR(S): Herz, H. G.; Kreuer, K. D.; Maier, J.; Scharfenberger,

G.; Schuster, M. F. H.; Meyer, W. H.

CORPORATE SOURCE: Max-Planck-Institut fur Festkorperforschung,

Stuttgart, D-70569, Germany

SOURCE: Electrochimica Acta (2003), 48(14-16), 2165-2171

CODEN: ELCAAV; ISSN: 0013-4686

PUBLISHER: Elsevier Science Ltd.

DOCUMENT TYPE: Journal LANGUAGE: English

AB The prepn. and characterization of fully polymeric-bound heterocycles as proton solvents is presented. Two different types of polymers are prepared: Polystyrene with imidazole terminated flexible side chains and benzimidazole covalently bonded to an inorg. SiO2 network by a flexible spacer. High proton conductivities of up to 7+10-4 S cm-1 at 200 °C have been obtained for these polymers in the absence of water. The presence of protonic charge carriers (protonated and deprotonated heterocycles) is a result of self-dissociation and the proton conductance is suggested to occur via intermol. proton transfer and structural reorganization within a dynamical hydrogen bond network formed by the aggregation of the terminating heterocycles.

IT 605658-12-6P

RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PNU (Preparation, unclassified); PRP (Properties); RCT (Reactant); PREP (Preparation); PROC (Process); RACT (Reactant or reagent) (fully polymeric proton solvents with high proton mobility on base of benzimidazole derivs. covalently bonded to SiO2)

RN 605658-12-6 HCAPLUS

CN Carbamic acid, [3-(triethoxysilyl)propyl]-, 5-(1H-benzimidazol-2-yl)pentyl ester (9CI) (CA INDEX NAME)

REFERENCE COUNT: 11 THERE ARE 11 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L10 ANSWER 14 OF 24 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2003:93943 HCAPLUS Full-text DOCUMENT NUMBER: 139:110787

TITLE: Enantioseparation of racemic N-acylarylalkylamines on

various amino alcohol derived π -acidic chiral

stationary phases

Ryoo, Jae Jeong; Kim, Tae Hyuk; Im, Sung Hyun; Jeong, AUTHOR(S):

Young Han; Park, Ji Yeon; Choi, Seong-Ho; Lee,

Kwang-Pill; Park, Jung Hag

CORPORATE SOURCE: Graduate School, Department of Chemistry, Kyungpook

> National University, Buk-Ku, Taegu, 702-701, S. Korea Journal of Chromatography, A (2003), 987(1-2), 429-438

SOURCE: CODEN: JCRAEY; ISSN: 0021-9673

Elsevier Science B.V. PUBLISHER:

DOCUMENT TYPE: Journal LANGUAGE: English

AΒ Five π -acidic chiral stationary phases (CSPs), CSP 4, CSP 5, CSP 6, CSP 7 and CSP 8, were prepared by connecting the N-(3,5-dimethylbenzoyl) derivative of (R)-alaninol, (S)-leucinol, (1S,2R)-ephedrine and (S)-tert-leucinol and the O-(3,5-dinitrobenzoyl) derivative of (R)-phenylglycinol to silica gel through a carbamate or urea linkage. The CSPs were applied to the resolution of various racemic N-acyl-1-naphthylaminoalkanes by chiral HPLC, and the chromatog. resolution results were compared with those of previously reported CSPs (CSP 2, CSP 3), which are derived from N-(3,5- dinitrobenzoyl)-(1S,2R)-norephedrine and N-(3,5-dinitrobenzoyl-(R)-phenylglycinol). Based on a comparison of the resolution results for each CSP, the role of each functional group on the five chiral selectors is explained.

ΤТ 445385-80-8DP, reaction product with silica gel

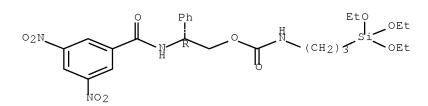
RL: ARU (Analytical role, unclassified); NUU (Other use, unclassified); SPN (Synthetic preparation); ANST (Analytical study); PREP (Preparation); USES (Uses)

(chiral stationary phase; enantiosepn. of racemic N-acylarylalkylamines on various amino alc. derived π -acidic chiral stationary phases)

445385-80-8 HCAPLUS RN

Carbamic acid, N-[3-(triethoxysilyl)propyl]-, (2R)-2-[(3,5-CN dinitrobenzoyl)amino]-2-phenylethyl ester (CA INDEX NAME)

Absolute stereochemistry.



REFERENCE COUNT: 16 THERE ARE 16 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L10 ANSWER 15 OF 24 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2003:93905 HCAPLUS Full-text

DOCUMENT NUMBER: 139:110784

TITLE: Comparison of monomeric and polymeric chiral

stationary phases

Lee, Kwang-Pill; Choi, Seong-Ho; Kim, Soo-Yeon; Kim, AUTHOR(S):

Tae-Hyuk; Ryoo, Jae Jeong; Ohta, Kazutoku; Jin, Ji-Ye;

Takeuchi, Toyohide; Fujimoto, Chuzo

Department of Chemistry Graduate School, Kyungpook CORPORATE SOURCE:

National University, Taegu, 702-701, S. Korea

Journal of Chromatography, A (2003), 987(1-2), 111-118SOURCE:

CODEN: JCRAEY; ISSN: 0021-9673

Elsevier Science B.V. PUBLISHER:

DOCUMENT TYPE: Journal LANGUAGE: English

Two-type polymeric chiral stationary phases (pCSPs) were prepd. by surface grafting of chiral acryl-type monomers on a silica gel surface modified with 3-(trimethoxysilyl)propylmethacrylate. The prepared pCSPs were characterized by IR, FT-Raman, SEM, and elemental anal. Two-type monomeric chiral stationary phases (mCSPs) were also prepared The racemic analytes were separated using the prepared mCSPs and pCSPs. The separation factor (α) and capacity factor (k1) of the racemic analytes for the pCSP and mCSP were compared. The α and k1 values of the mCSP were higher than those of the pCSP. IT 445385-80-80P, reaction product with silica gel

RL: ARU (Analytical role, unclassified); NUU (Other use, unclassified); PRP (Properties); SPN (Synthetic preparation); ANST (Analytical study); PREP (Preparation); USES (Uses)

(monomeric chiral stationary phase; preparation and comparison of monomeric and polymeric chiral stationary phases for HPLC)

RN 445385-80-8 HCAPLUS

CN Carbamic acid, N-[3-(triethoxysily1)propy1]-, (2R)-2-[(3,5-dinitrobenzoy1)amino]-2-phenylethyl ester (CA INDEX NAME)

Absolute stereochemistry.

IT 445385-80-8P

RL: ARU (Analytical role, unclassified); PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); ANST (Analytical study); PREP (Preparation); RACT (Reactant or reagent)

(preparation and comparison of monomeric and polymeric chiral stationary phases for HPLC)

RN 445385-80-8 HCAPLUS

CN Carbamic acid, N-[3-(triethoxysilyl)propyl]-, (2R)-2-[(3,5-dinitrobenzoyl)amino]-2-phenylethyl ester (CA INDEX NAME)

Absolute stereochemistry.

REFERENCE COUNT: 15 THERE ARE 15 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L10 ANSWER 16 OF 24 USPATFULL on STN

ACCESSION NUMBER: 2002:95848 USPATFULL Full-text

TITLE: Amorphous silica having discrete voids and

spatially organized functionalities formed therein

INVENTOR(S): Katz, Alexander, Flagstaff, AZ, United States

Davis, Mark E., Pasadena, CA, United States

PATENT ASSIGNEE(S): California Institute of Technology, Pasadena, CA,

United States (U.S. corporation)

NUMBER KIND DATE US 6380266 B1 20020430 US 1998-96991 19980612 (9)

PATENT INFORMATION:
APPLICATION INFO.:

NUMBER DATE _____

PRIORITY INFORMATION: US 1997-49615P 19970613 (60)

DOCUMENT TYPE: Utility

FILE SEGMENT: GRANTED
PRIMARY EXAMINER: Lovering, Richard D. LEGAL REPRESENTATIVE: Pillsbury Winthrop LLP

NUMBER OF CLAIMS: 40 EXEMPLARY CLAIM:

NUMBER OF DRAWINGS: 17 Drawing Figure(s); 14 Drawing Page(s) LINE COUNT: 1316

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

The present invention relates to amorphous inorganic materials having pores of controlled size and shape with one or more spatially organized functional groups formed therein. Much like the active site residues in an enzyme, these functional groups can be positioned in a defined three dimensional relationship within the voids and with respect to each other. By varying both the positions and identities of these functional groups, diverse sets of substrate specific adsorbents and non-biologically-based catalysts can be made.

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

IT 218155-73-8D, reaction products with silica

(one-point imprint structure, silica-supported; preparation and properties of silica gel and silica with voids containing spatially organized organic functionalities)

RN 218155-73-8 USPATFULL

CN Carbamic acid, [3-(triethoxysilyl)propyl]-, phenylmethyl ester (9CI) (CA INDEX NAME)

L10 ANSWER 17 OF 24 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2002:331556 HCAPLUS Full-text

DOCUMENT NUMBER: 137:162940

TITLE: Dendritic polymers as bonded stationary phases in

capillary electrochromatography

AUTHOR(S): Chao, Helen C.; Hanson, James E.
CORPORATE SOURCE: Johnson and Johnson Pharmaceutical Research and

Development, LLC, Raritan, NJ, 08869, USA

Journal of Separation Science (2002), 25(5/6), 345-350 SOURCE:

CODEN: JSSCCJ; ISSN: 1615-9306

PUBLISHER: Wiley-VCH Verlag GmbH DOCUMENT TYPE: Journal LANGUAGE: English

Dendrimers are an interesting new class of polymers. Their unusual chem. and phys. properties have great potential in many applications. Polyamidoamine starburst dendrimers (PAMAM-SBDs) showed promising results as a pseudostationary phase in Micellar Electrokinetic Capillary Chromatog. (MECC). However, few chromatog. applications were explored using dendrimers. The objective of the authors' study is to explore the potential of using poly(aryl ether) monodendrons as a bonded stationary phase in open tubular Capillary Electrochromatog. (CEC). Organic materials were bonded to the interior of silica capillaries through a triethoxy-(3-isocyanatopropyl)silane linker. Bonded materials included GO, G1, and G2 poly(aryl ether) monodendrons and C18 alkyl chains. The bonded materials tended to reduce the EOF, and showed promise in a number of test sepns., including neutral aromatic hydrocarbons and basic proteins. This method shows promise for the development of new CEC methods for separation science.

IT 218155-73-8

RL: ARU (Analytical role, unclassified); NUU (Other use, unclassified); ANST (Analytical study); USES (Uses)

(coating material; dendritic polymers as bonded stationary phases in capillary electrochromatog.)

RN 218155-73-8 HCAPLUS

CN Carbamic acid, [3-(triethoxysily1)propy1]-, phenylmethyl ester (9CI) (CA INDEX NAME)

REFERENCE COUNT: 21 THERE ARE 21 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L10 ANSWER 18 OF 24 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2002:322688 HCAPLUS Full-text

DOCUMENT NUMBER: 137:162939

TITLE: NMR studies of chiral discrimination relevant to the

enantioseparation of N-acylarylalkylamines by an

(R)-phenylglycinol-derived chiral selector

AUTHOR(S): Im, Sung Hyun; Ryoo, Jae Jeong; Lee, Kwang-Pill; Choi,

Seong-Ho; Jeong, Young Han; Jung, Young Sik; Hyun,

Myung Ho

CORPORATE SOURCE: Analytical Laboratory, R and D Institute, LG Household

and Health Care, Taejon, S. Korea Chirality (2002), 14(4), 329-333

CODEN: CHRLEP; ISSN: 0899-0042

PUBLISHER: Wiley-Liss, Inc.

DOCUMENT TYPE: Journal LANGUAGE: English

SOURCE:

AB Recently, it is reported that the chiral recognition ability of (R)-N-3,5-dinitrobenzoyl phenylglycinol derivative was examined as a new HPLC chiral stationary phase (CSP 1) for the resolution of racemic N-acylnaphthylalkylamines. However, the mechanism of chiral discrimination on the CSP remained elusive until now. A spectroscopic study of the chiral discrimination mechanism of CSP 1 was undertaken using mixts. of (R)-N-3,5-dinitrobenzoyl phenylglycinol-derived chiral selector (2) and each of the enantiomers of N-acylnaphthylalkylamines (3) by NMR study. First, the

differences in free energy changes ($\Delta\Delta G$) upon diastereomeric complexation in solution between the complex of each isomer with chiral selector 2 by NMR titration were calculated. The values were then compared with those estimated by chiral HPLC. The chemical shift changes of each proton on the chiral selector and analytes were also checked and the chemical shift changes decreased continuously as the acyl group on analytes increased in length. This observation was consistent with the HPLC data. From these exptl. results, the interaction mechanism of chiral discrimination between the chiral selector and the analytes is more precisely explained.

IT 445385-80-8DP, reaction product with silica gel

RL: ARU (Analytical role, unclassified); NUU (Other use, unclassified); SPN (Synthetic preparation); ANST (Analytical study); PREP (Preparation); USES (Uses)

(NMR studies of chiral discrimination relevant to the enantiosepn. of N-acylarylalkylamines by an (R)-phenylglycinol-derived chiral selector) 445385-80-8 HCAPLUS

CN Carbamic acid, N-[3-(triethoxysilyl)propyl]-, (2R)-2-[(3,5-dinitrobenzoyl)amino]-2-phenylethyl ester (CA INDEX NAME)

Absolute stereochemistry.

RN

REFERENCE COUNT: 19 THERE ARE 19 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L10 ANSWER 19 OF 24 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2000:566169 HCAPLUS Full-text

DOCUMENT NUMBER: 133:305115

TITLE: Dendrimers as stationary phases in capillary

electrochromatography

AUTHOR(S): Chao, Helen; Hanson, James E.

CORPORATE SOURCE: Department of Chemistry, Seton Hall University, South

Orange, NJ, 07079, USA

SOURCE: Polymeric Materials Science and Engineering (2000),

83, 438-439

CODEN: PMSEDG; ISSN: 0743-0515

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal LANGUAGE: English

This study is focused on open tubular capillary electrochromatog. (OT-CEC) in which a stationary phase is bonded inside a capillary tube by a covalent bond. The first generation (GI) and second generation (G2) of alc. focused poly(aryl ether) monodendrons were used. A method to chemical bond the dendritic polymers to the inner wall of fused silica capillary columns was developed with 3-(triethoxysilyl)propyl isocyanate as anchoring agent. Carbamates from four different alcs. were studied as bonded stationary phases in OT-CEC: (1) GI alc., (2) G2 alc., (3) stearyl alc. (C18 linear), and (4) 4-hydroxymethyl pyridine. Model sepns. of different analytes were examined over a range of pH values and a variety of buffer electrolytes. It is the authors' hypothesis that the permanent silanol modification should provide better control of EOF

and prevent undesired analyte-wall interactions arising from electrostatic attractions.

IT 218155-73-8

RL: ARU (Analytical role, unclassified); RCT (Reactant); ANST (Analytical study); RACT (Reactant or reagent)

(dendrimers as stationary phases in capillary electrochromatog.)

RN 218155-73-8 HCAPLUS

CN Carbamic acid, [3-(triethoxysily1)propy1]-, phenylmethyl ester (9CI) (CA INDEX NAME)

REFERENCE COUNT: 21 THERE ARE 21 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L10 ANSWER 20 OF 24 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2000:76414 HCAPLUS Full-text

DOCUMENT NUMBER: 132:222568

TITLE: Molecular imprinting of bulk, microporous

silica

AUTHOR(S): Katz, Alexander; Davis, Mark E.

CORPORATE SOURCE: Chemical Engineering, California Institute of

Technology, Pasadena, CA, 91125, USA

SOURCE: Nature (London) (2000), 403(6767), 286-289

CODEN: NATUAS; ISSN: 0028-0836

PUBLISHER: Nature Publishing Group

DOCUMENT TYPE: Journal LANGUAGE: English

Mol. imprinting aims to create solid materials contq. chem. functionalities AB that are spatially organized by covalent or noncovalent interactions with imprint (or template) mols. during the synthesis process. Subsequent removal of the imprint mols. leaves behind designed sites for the recognition of small mols., making the material ideally suited for applications such as sepns., chemical sensing and catalysis. Until now, the mol. imprinting of bulk polymers and polymer and SiO2 surfaces was reported, but the extension of these methods to a wider range of materials remains problematic. For example, the formation of substrate-specific cavities within bulk SiO2, while conceptually straightforward, was difficult to accomplish exptl. Here the authors describe the imprinting of bulk amorphous silicas with single aromatic rings carrying up to three 3-aminopropyltriethoxysilane side groups; this generates and occupies microporosity and attaches functional organic groups to the pore walls in a controlled fashion. The triethoxysilane part of the mols. side groups is incorporated into the SiO2 framework during sol-gel synthesis, and subsequent removal of the aromatic core creates a cavity with spatially organized aminopropyl groups covalently anchored to the pore walls. The imprinted silicas act as shape-selective base catalysts. Strategy can be extended to imprint other functional groups, which should give access to a wide range of functionalized materials.

IT 218155-73-8P 261165-60-0P

RL: CAT (Catalyst use); RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent); USES (Uses)

(sol-gel polymerization of tetraethoxysilane using; mol. imprinting of bulk,

microporous silica)

RN 218155-73-8 HCAPLUS

CN Carbamic acid, [3-(triethoxysily1)propy1]-, phenylmethyl ester (9CI) (CA INDEX NAME)

RN 261165-60-0 HCAPLUS

CN Carbamic acid, [3-(triethoxysilyl)propyl]-, 1,4-phenylenebis(methylene) ester (9CI) (CA INDEX NAME)

REFERENCE COUNT: 18 THERE ARE 18 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L10 ANSWER 21 OF 24 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1999:7889 HCAPLUS Full-text

DOCUMENT NUMBER: 130:68593

TITLE: Preparation and properties of silica gel and

silica with voids containing spatially organized

organic functionalities

INVENTOR(S): Katz, Alexander; Davis, Mark E.

PATENT ASSIGNEE(S): California Institute of Technology, USA

SOURCE: PCT Int. Appl., 69 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

	PATENT NO.					KIND DATE				APPLICATION NO.						DATE			
	WO	9856	 498			A1		1998:	1217	,	 WO 1	998-	JS12:	352		1:	9980	612	
		W:	AL,	AM,	AT,	AU,	ΑZ,	BA,	BB,	BG,	BR,	BY,	CA,	CH,	CN,	CU,	CZ,	DE,	
			DK,	EE,	ES,	FΙ,	GB,	GE,	GH,	GM,	GW,	HU,	ID,	IL,	IS,	JP,	ΚE,	KG,	
			KP,	KR,	KΖ,	LC,	LK,	LR,	LS,	LT,	LU,	LV,	MD,	MG,	MK,	MN,	MW,	MX,	
			NO,	NZ,	PL,	PT,	RO,	RU,	SD,	SE,	SG,	SI,	SK,	SL,	ΤJ,	TM,	TR,	TT,	
			UA,	UG,	UZ,	VN,	YU,	ZW											
		RW:	GH,	GM,	KΕ,	LS,	MW,	SD,	SZ,	UG,	ZW,	ΑT,	BE,	CH,	CY,	DE,	DK,	ES,	
			FI,	FR,	GB,	GR,	IE,	ΙΤ,	LU,	MC,	NL,	PT,	SE,	BF,	ВJ,	CF,	CG,	CI,	
			CM,	GΑ,	GN,	ML ,	MR,	NE,	SN,	TD,	ΤG								
	ΑU	9879	656			Α		1998	1230		AU 1	998-	7965	6		1	9980	612	
	US	6380	266			В1		2002	0430		US 1	998-	9699:	1		1	9980	612	
PRIC	DRITY	APP:	LN.	INFO	.:						US 1	997-	4961.	5P]	P 19	9970	613	
										,	WO 1	998-1	JS12:	352	Ī	W 1	9980	612	

AB Porous silica gel and amorphous silica having voids containing spatially organized organic functionalities were prepared such that the organic functionalities were located in similar positions in each void and are homogeneously distributed. The organic functionalities are bound covalently to the available Si sites by a number of functional groups and can contain a number of functional groups themselves, including transition metal complexes, metal ions, etc. These functional groups can be positioned in a defined three-dimensional relationship within the voids and with respect to each other, possibly even to selectively position a chiral mol. imprinting structure within the void. By varying both the positions and identities of these functional groups, potentially diverse sets of substrate-specific adsorbents and non-biol.-based catalysts can be made.

IT 218155-73-8D, reaction products with silica

RL: RCT (Reactant); TEM (Technical or engineered material use); RACT (Reactant or reagent); USES (Uses)

(one-point imprint structure, silica-supported; preparation and properties of silica gel and silica with voids containing spatially organized organic functionalities)

RN 218155-73-8 HCAPLUS

CN Carbamic acid, [3-(triethoxysily1)propy1]-, phenylmethyl ester (9CI) (CA INDEX NAME)

REFERENCE COUNT: 9 THERE ARE 9 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L10 ANSWER 22 OF 24 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1998:45539 HCAPLUS Full-text

DOCUMENT NUMBER: 128:186151

TITLE: Thermal stability of NLO sol-qel networks with

reactive chromophores

AUTHOR(S): Sung, Po-Hou; Hsu, Tien-Fong

CORPORATE SOURCE: Department of Chemistry, Fu-Jen Catholic University,

Taipei, Taiwan

SOURCE: Polymer (1998), 39(6-7), 1453-1459

CODEN: POLMAG; ISSN: 0032-3861

PUBLISHER: Elsevier Science Ltd.

DOCUMENT TYPE: Journal LANGUAGE: English

AB Transparent SiO2 films with org. chromophores as side chain, main chain or crosslinking agent were prepared by the sol-gel process. The resulting films heated at $100-120^{\circ}$ under high corona field exhibited d33 = 16-29 pm V-1 in Maker-fringe measurement. The crosslinking type chromophore exhibited optimum poling orientation stability.

IT 203209-89-6

RL: PRP (Properties)

(thermal stability of NLO sol-gel networks with reactive chromophores)

RN 203209-89-6 HCAPLUS

CN Carbamic acid, [3-(triethoxysilyl)propyl]-, [[4-[[2-[(9,9-diethoxy-4-oxo-3,10-dioxa-5-aza-9-siladodec-1-yl)oxy]-4-nitrophenyl]azo]phenyl]imino]di-2,1-ethanediyl ester (9CI) (CA INDEX NAME)

PAGE 1-B

REFERENCE COUNT: 14 THERE ARE 14 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L10 ANSWER 23 OF 24 USPATFULL on STN

ACCESSION NUMBER: 97:38570 USPATFULL Full-text

TITLE: Ultraviolet absorber and coating material

INVENTOR(S): Suga, Masanobu, Yokohama, Japan Asano, Tsuyoshi, Yokohama, Japan

Kuroda, Nobuyuki, Yokohama, Japan

PATENT ASSIGNEE(S): Nippon Oil Co., Ltd., Tokyo, Japan (non-U.S.

corporation)

NUMBER DATE

PRIORITY INFORMATION: JP 1995-52992 19950313

DOCUMENT TYPE: Utility FILE SEGMENT: Granted

PRIMARY EXAMINER: Hoke, Veronica P. LEGAL REPRESENTATIVE: Darby & Darby

NUMBER OF CLAIMS: 19 EXEMPLARY CLAIM: 1

NUMBER OF DRAWINGS: 8 Drawing Figure(s); 2 Drawing Page(s)

LINE COUNT: 661

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

An ultraviolet absorber contains as an effective component a compound represented by the formula: ##STR1## wherein R.sup.1 denotes a hydrogen atom, a halogen atom or an alkyl group having 1 to 10 carbon atoms, R.sup.2 denotes a hydrogen atom or an alkyl group having 1 to 10 carbon atoms, R.sup.3 and R.sup.4 denote the same or different groups and denote an alkylene group having 1 to 10 carbon atoms, R.sup.5 to R.sup.9 denote the

same or different groups and denote an alkyl group having 1 to 10 carbon atoms, an alkoxy group having 1 to 10 carbon atoms, and an aryl group having 6 to 10 carbon atoms, a hydroxyl group or a hydrogen atom, X denotes an amido bond (CONH), a urethane bond (OCONH) or an ester bond (COO), and n is an integer of $n \ge 0$. A coating material contains the ultraviolet absorber.

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

IT 182806-37-7P

(UV absorber and coating material)

RN 182806-37-7 USPATFULL

CN Carbamic acid, [3-(triethoxysily1)propy1]-, 2-[3-(2H-benzotriazol-2-y1)-4-hydroxyphenyl]ethyl ester (9CI) (CA INDEX NAME)

L10 ANSWER 24 OF 24 HCAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: 1994:65979 HCAPLUS Full-text

DOCUMENT NUMBER: 120:65979

TITLE: Manufacture of photochromic hybrid silida

gels

INVENTOR(S): Matsuda, Reiko; Yamaguchi, Masao; Okamoto, Hidenori

PATENT ASSIGNEE(S): Tokuyama Soda Kk, Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 9 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 05202190	A	19930810	JP 1992-12394	19920127
JP 3187907	B2	20010716		
PRIORITY APPLN. INFO.:			JP 1992-12394	19920127

AB Photochromic silica gels are manufd. by (A) the reaction of alkoxysilanes Si(OR1)4 (I: R1 = H, alkyl) or R2nSi(OR3)4-n (II: R2 = alkyl, alkenyl, Ph; R3 = H, alkyl; n = 1, 2) with alkoxysilyl group-containing photochromic compds. and treating the resulting silica gels with M(OR)n (III: R = H, alkyl; M = alkali metal, alkaline earth metal; n = valence of metal) or (B) the reaction of I or II with the photochromic compds. in the presence of III. Acid strength in the silica gels is controlled and hence the photochromic characteristics are improved.

IT 152238-73-8

RL: RCT (Reactant); RACT (Reactant or reagent)
 (reaction of, with alkoxysilanes, photochromic silica gels
 from)

RN 152238-73-8 HCAPLUS

CN Carbamic acid, [3-(triethoxysily1)propy1]-, 2-(3,3-dimethy1-8'-

nitrospiro[2H-indole-2,3'-[3H]naphth[2,1-b][1,4]oxazin]-1(3H)-y1)ethyl
ester (9CI) (CA INDEX NAME)

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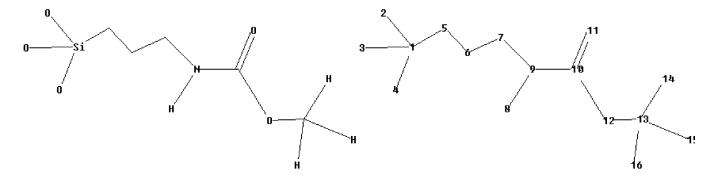
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http://www.cas.org/support/stngen/stndoc/properties.html

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chain nodes :

1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16

chain bonds :

 $1-2 \quad 1-3 \quad 1-4 \quad 1-5 \quad 5-6 \quad 6-7 \quad 7-9 \quad 8-9 \quad 9-10 \quad 10-11 \quad 10-12 \quad 12-13 \quad 13-14 \quad 13-15$

13-16

exact/norm bonds :

7-9 9-10 10-11 10-12 12-13

exact bonds :

1-2 1-3 1-4 1-5 5-6 6-7 8-9 13-14 13-15 13-16

Match level :

1:CLASS 2:CLASS 3:CLASS 4:CLASS 5:CLASS 6:CLASS 7:CLASS 8:CLASS 9:CLASS 10:CLASS 11:CLASS 12:CLASS 13:CLASS 14:CLASS 15:CLASS 16:CLASS

L11 STRUCTURE UPLOADED

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SAMPLE SEARCH INITIATED 07:42:02 FILE 'REGISTRY' SAMPLE SCREEN SEARCH COMPLETED - 115 TO ITERATE

100.0% PROCESSED 115 ITERATIONS 0 ANSWERS

SEARCH TIME: 00.00.01

FULL FILE PROJECTIONS: ONLINE **COMPLETE** BATCH **COMPLETE**

PROJECTED ITERATIONS: 1657 TO 2943 0 TO PROJECTED ANSWERS:

L12 0 SEA SSS SAM L11

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FULL SEARCH INITIATED 07:42:08 FILE 'REGISTRY' FULL SCREEN SEARCH COMPLETED - 2077 TO ITERATE

100.0% PROCESSED 2077 ITERATIONS 11 ANSWERS

TOTAL

SEARCH TIME: 00.00.01

L13 11 SEA SSS FUL L11

=> FIL HCAPLUS USPATFULL BIOSIS MEDLINE COST IN U.S. DOLLARS

SINCE FILE ENTRY SESSION

FULL ESTIMATED COST 178.36 573.11

DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS) SINCE FILE TOTAL ENTRY SESSION 0.00 -20.80

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FILE 'MEDLINE' ENTERED AT 07:42:15 ON 14 MAR 2008

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PROCESSING COMPLETED FOR L15

L16 7 DUP REM L15 (0 DUPLICATES REMOVED)

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L16 ANSWER 1 OF 7 USPATFULL on STN

ACCESSION NUMBER: 2006:289411 USPATFULL Full-text

TITLE: Polymers with urea groups and silyl groups and

production and use thereof

INVENTOR(S): Bachon, Thomas, Duesseldorf, GERMANY, FEDERAL REPUBLIC

Huebner, Wilfried, Langenfeld, GERMANY, FEDERAL

REPUBLIC OF

Kluth, Hermann, Duesseldorf, GERMANY, FEDERAL REPUBLIC

Klauck, Wolfgang, Meerbusch, GERMANY, FEDERAL REPUBLIC

Klein, Johann, Duesseldorf, GERMANY, FEDERAL REPUBLIC

PATENT ASSIGNEE(S): Henkel Kommanditgesellschaft Auf Aktien (Henkel KGAA)

(non-U.S. corporation)

NUMBER KIND DATE ______ US 2006247407 A1 20061102 US 2006-360073 A1 20060223 (11) PATENT INFORMATION: APPLICATION INFO.:

RELATED APPLN. INFO.: Division of Ser. No. US 2003-415185, filed on 25 Apr

2003, GRANTED, Pat. No. US 7057001 A 371 of

International Ser. No. WO 2001-EP12291, filed on 24 Oct

2001

NUMBER DATE

PRIORITY INFORMATION: DE 2000-10053545 20001027

DOCUMENT TYPE: Utility FILE SEGMENT: APPLICATION LEGAL REPRESENTATIVE: WOODCOCK WASHBURN LLP, ONE LIBERTY PLACE, 46TH FLOOR,

PHILADELPHIA, PA, 19103, US

NUMBER OF CLAIMS: 20 EXEMPLARY CLAIM: 1 LINE COUNT: 1473

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB A process for the production of compounds which contain at least one urea group and at least one silyl goup. The process involves reacting a compound containing at least one amino group and a carbamate compound. The compounds produced by this process are useful as surface coating compositions, foams, and adhesives.

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

IT 23432-62-4DP, polymers with dimer acid polyamides

419572-67-1P 419572-68-2P 419572-69-3P

420088-12-6P

(production and use of polymers with urea and silyl groups)

RN 23432-62-4 USPATFULL

CN Carbamic acid, N-[3-(trimethoxysilyl)propyl]-, methyl ester (CA INDEX NAME)

RN 419572-67-1 USPATFULL

CN Carbamic acid, [3-(trimethoxysilyl)propyl]-, methyl ester, polymer with 1-octanamine (9CI) (CA INDEX NAME)

CM 1

CRN 23432-62-4 CMF C8 H19 N O5 Si

CM 2

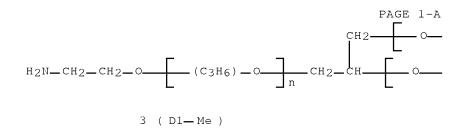
CRN 111-86-4 CMF C8 H19 N

 H_2N — (CH2)7—Me

CN Carbamic acid, [3-(trimethoxysilyl)propyl]-, methyl ester, polymer with $\alpha,\alpha',\alpha''-1$,2,3-propanetriyltris[ω -(2-aminomethylethoxy)poly[oxy(methyl-1,2-ethanediyl)]] (9CI) (CA INDEX NAME)

CM 1

CRN 64852-22-8 CMF (C3 H6 O)n (C3 H6 O)n (C3 H6 O)n C12 H29 N3 O3 CCI IDS, PMS CDES *



PAGE 1-B

CM 2

CRN 23432-62-4 CMF C8 H19 N O5 Si

RN 419572-69-3 USPATFULL

CN Carbamic acid, [3-(trimethoxysilyl)propyl]-, methyl ester, polymer with $\alpha\text{-(2-aminomethylethyl)}-\omega\text{-(2-aminomethylethoxy)}poly[oxy(methyl-1,2-ethanediyl)] (9CI) (CA INDEX NAME)$

CM 1

CRN 23432-62-4 CMF C8 H19 N O5 Si

$$\begin{array}{c} \text{OMe} \\ \text{MeO-Si-(CH2)_3-NH-C-OMe} \\ \\ \text{OMe} \end{array}$$

CM 2

CRN 9046-10-0

CMF (C3 H6 O)n C6 H16 N2 O

CCI IDS, PMS

CDES *

2 (D1—Me)

RN 420088-12-6 USPATFULL

CN Carbamic acid, [3-(trimethoxysilyl)propyl]-, methyl ester, polymer with Macromelt TPX 22-405 (9CI) (CA INDEX NAME)

CM 1

CRN 420088-11-5 CMF Unspecified CCI PMS, MAN

STRUCTURE DIAGRAM IS NOT AVAILABLE

CM 2

CRN 23432-62-4 CMF C8 H19 N O5 Si

L16 ANSWER 2 OF 7 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2005:1319723 HCAPLUS Full-text

DOCUMENT NUMBER: 144:219440

TITLE: Preparation and evaluation of novel stationary phases

for improved chromatographic purification of

pneumocandin B0

AUTHOR(S): Welch, Christopher J.; DaSilva, Jimmy O.; Nti-Gyabaah,

Joseph; Antia, Firoz; Goklen, Kent; Boyd, Russell

CORPORATE SOURCE: Merck Research Laboratories, Merck & Co. Inc., Rahway,

NJ, 07065, USA

SOURCE: Journal of Chromatography, A (2006), 1101(1-2),

CODEN: JCRAEY; ISSN: 0021-9673

PUBLISHER: Elsevier B.V.

DOCUMENT TYPE: Journal LANGUAGE: English

Prepn. and evaluation of a no. of stationary phases for improved chromatog. purification of pneumocandin BO, a key intermediate in the synthesis of the antifungal agent, Cancidas, has led to the identification of several materials with potential for improved performance.

26411-01-8 ΙT

RL: AMX (Analytical matrix); ANST (Analytical study)

(preparation and evaluation of novel stationary phases for improved

chromatog. purification of pneumocandin B0)

RN 26411-01-8 HCAPLUS

CN Carbamic acid, N-[3-(triethoxysilyl)propyl]-, methyl ester (CA INDEX NAME)

17 THERE ARE 17 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L16 ANSWER 3 OF 7 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2005:540471 HCAPLUS Full-text

DOCUMENT NUMBER: 143:78642

TITLE: Method for producing isocyanate-organosilanes in

presence of fluidized solid particles

Ruedinger, Christoph; Eberle, Hans-Juergen INVENTOR(S):

PATENT ASSIGNEE(S): Consortium fuer Elektrochemische Industrie G.m.b.H.,

Germany

SOURCE: PCT Int. Appl., 32 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: German

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.				KIND DATE			1	APPLICATION NO.						DATE				
WO 2005055974 WO 2005055974				A2 A3		20050623 20050728		1	WO 2004-EP13722						20041202			
	W: RW:	CN, GE, LK, NO, TJ, BW, AZ, EE,	CO, GH, LR, NZ, TM, GH, BY, ES, SE,	CR, GM, LS, OM, TN, GM, KG, FI,	CU, HR, LT, PG, TR, KE, KZ,	CZ, HU, LU, PH, TT, LS, MD, GB, TR,	AU, DE, ID, LV, PL, TZ, MW, RU, GR, BF,	DK, IL, MA, PT, UA, MZ, TJ, HU,	DM, IN, MD, RO, UG, NA, TM, IE,	DZ, IS, MG, RU, US, SD, AT, IS,	EC, JP, MK, SC, UZ, SL, BE, IT,	EE, KE, MN, SD, VC, SZ, BG, LT,	EG, KG, MW, SE, VN, TZ, CH, LU,	ES, KP, MX, SG, YU, UG, CY, MC,	FI, KR, MZ, SK, ZA, ZM, CZ, NL,	GB, KZ, NA, SL, ZM, ZW, DE,	GD, LC, NI, SY, ZW AM, DK, PT,	

DE	10358	8064			A1		2005	0714	DI	Ξ 2	003 - 3	1035	8064		2	0031	211
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	R:	ΑT,	BE,	CH,	DE,	DK,	ES,	FR,	GB, (GR,	ΙΤ,	LI,	LU,	NL,	SE,	MC,	PT,
		ΙE,	SI,	LT,	FΙ,	RO,	CY,	TR,	BG, (CZ,	EE,	HU,	PL,	SK,	IS		
CN	18393	140			А	,	2006	0927	CI	1 2	004-	8002	4229		2	0041	202
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AB The invention relates to a method for producing isocyanate-organosilanes by thermolysis of carbamate-organosilanes in the presence of solid fluidized particles which may included catalysts. Thus, γ - isocyanatopropyltrimethoxysilane prepared by heterogeneous-catalyzed thermolysis of methylcarbamatopropyltrimethoxysilane in a fluidized bed of iron oxide showed a yield of 75% compared to 56% for a similar process without the iron oxide fluidized bed.

IT 23432-62-4

RL: RCT (Reactant); RACT (Reactant or reagent) (producing isocyanate-organosilanes in presence of fluidized solid particles)

RN 23432-62-4 HCAPLUS

CN Carbamic acid, N-[3-(trimethoxysily1)propy1]-, methyl ester (CA INDEX NAME)

L16 ANSWER 4 OF 7 USPATFULL on STN

ACCESSION NUMBER: 2005:298736 USPATFULL Full-text

TITLE: Polymers with urea groups and silyl groups and

production and use thereof

INVENTOR(S): Bachon, Thomas, Duesseldorf, GERMANY, FEDERAL REPUBLIC

OF

Huebner, Wilfried, Langenfeld, GERMANY, FEDERAL

REPUBLIC OF

Kluth, Hermann, Duesseldorf, GERMANY, FEDERAL REPUBLIC

OF

Klauck, Wolfgang, Meersbusch, GERMANY, FEDERAL REPUBLIC

OF

Klein, Johann, Duesseldorf, GERMANY, FEDERAL REPUBLIC

OF

	NUMBER	KIND	DATE	
PATENT INFORMATION:	US 2005260401	A1	20051124	
	US 7057001	В2	20060606	
APPLICATION INFO.:	US 2003-415185	A1	20011024	(10)
	WO 2001-EP12291		20011024	
			20030425	PCT 371 date

NUMBER	DATE						

PRIORITY INFORMATION: DE 2000-10053545 20001027

DOCUMENT TYPE: Utility FILE SEGMENT: APPLICATION

LEGAL REPRESENTATIVE: WOODCOCK WASHBURN LLP,, ONE LIBERTY PLACE, 46TH FLOOR,

PHILADELPHIA, PA, 19103, US

NUMBER OF CLAIMS: 17 EXEMPLARY CLAIM: 1-20 LINE COUNT: 1432

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB A process for the production of compounds which contain at least one urea group and at least one silyl group. The process involves reacting a compound containing at least one amino group and a carbamate compound. The compounds produced by this process are useful as surface coating compositions, foams and adhesives.

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

IT 23432-62-4DP, polymers with dimer acid polyamides

419572-67-1P 419572-68-2P 419572-69-3P

420088-12-6P

(production and use of polymers with urea and silyl groups)

RN 23432-62-4 USPATFULL

CN Carbamic acid, N-[3-(trimethoxysilyl)propyl]-, methyl ester (CA INDEX NAME)

RN 419572-67-1 USPATFULL

CN Carbamic acid, [3-(trimethoxysilyl)propyl]-, methyl ester, polymer with 1-octanamine (9CI) (CA INDEX NAME)

CM 1

CRN 23432-62-4 CMF C8 H19 N O5 Si

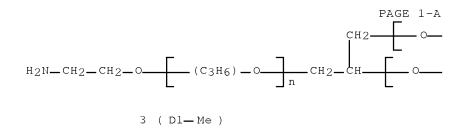
CM 2

CRN 111-86-4 CMF C8 H19 N RN 419572-68-2 USPATFULL

CN Carbamic acid, [3-(trimethoxysilyl)propyl]-, methyl ester, polymer with $\alpha, \alpha', \alpha''-1, 2, 3$ -propanetriyltris[ω -(2- aminomethylethoxy)poly[oxy(methyl-1,2-ethanediyl)]] (9CI) (CA INDEX NAME)

CM 1

CRN 64852-22-8 CMF (C3 H6 O)n (C3 H6 O)n (C3 H6 O)n C12 H29 N3 O3 CCI IDS, PMS CDES *



PAGE 1-B

CM 2
CRN 23432-62-4

CMF C8 H19 N O5 Si

RN 419572-69-3 USPATFULL

CN Carbamic acid, [3-(trimethoxysilyl)propyl]-, methyl ester, polymer with $\alpha - (2-\text{aminomethylethyl}) - \omega - (2-\text{aminomethylethoxy}) \text{poly}[\text{oxy}(\text{methylethyl}) - (2-\text{aminomethylethoxy})]$ (CA INDEX NAME)

CM 1

CRN 23432-62-4 CMF C8 H19 N O5 Si

CM 2

CRN 9046-10-0

CMF (C3 H6 O)n C6 H16 N2 O

CCI IDS, PMS

CDES *

2 (D1—Me)

RN 420088-12-6 USPATFULL

CN Carbamic acid, [3-(trimethoxysilyl)propyl]-, methyl ester, polymer with Macromelt TPX 22-405 (9CI) (CA INDEX NAME)

CM 1

CRN 420088-11-5

CMF Unspecified

CCI PMS, MAN

STRUCTURE DIAGRAM IS NOT AVAILABLE

CM 2

CRN 23432-62-4

CMF C8 H19 N O5 Si

L16 ANSWER 5 OF 7 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1999:798978 HCAPLUS Full-text

DOCUMENT NUMBER: 132:126367

TITLE: Composition-structure relations in organically

modified silica gels

AUTHOR(S): Husing, Nicola; Schubert, Ulrich

CORPORATE SOURCE: Institut fur Anorganische Chemie, Technische

Universitat Wien, Vienna, A-1060, Austria

Materials Research Society Symposium Proceedings

(1999), 576 (Organic/Inorganic Hybrid Materials II),

117-127

CODEN: MRSPDH; ISSN: 0272-9172

PUBLISHER: Materials Research Society

DOCUMENT TYPE: Journal LANGUAGE: English

SOURCE:

Monolithic silica aerogels modified by functional org, groups were prepared by base-catalyzed sol-gel processing of Si(OR)4/R'Si(OMe)3 mixts. (R' = organofunctional group), followed by drying of the wet gels with supercrit. CO2. When the functional organic group has only weakly or no basic properties, the microstructure of the obtained aerogels is similar to that of an unmodified silica aerogel prepared under the same conditions and quite independent of the kind of functional group. The exptl. findings are explained by a two-stage process in which the R'Si.tplbond. units condense to a pre-formed gel network obtained by hydrolysis and condensation of Si(OR)4. An increasing portion of R'Si(OMe)3 has the same effects on the hydrolysis and condensation reactions as decreasing the bulk d. of an unmodified silica aerogel and the same structural consequences as increasing the water/silane ratio and the catalyst concentration acting on Si(OR)4. This leads to larger primary particles and, associated with that, smaller sp. surface areas. The two-stage process is not observed when R' contains a strongly basic substituent such as NH2 or NHCH2CH2NH2. The structural parameters indicate that in these cases both R'Si(OMe)3 and Si(OR)4 are involved in the build-up of the gel network. This can be explained by extensive hydrogen bonding between the amino groups and silanol groups.

256342-59-3P ΤТ

> RL: PEP (Physical, engineering or chemical process); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); PROC (Process)

(ceramers; composition-structure relations in organically modified silica aerogels)

RN 256342-59-3 HCAPLUS

CN Carbamic acid, [3-(trimethoxysilyl)propyl]-, methyl ester, polymer with silicic acid (H4SiO4) tetramethyl ester (9CI) (CA INDEX NAME)

CM 1

CRN 23432-62-4 CMF C8 H19 N O5 Si

CM 2

CRN 681-84-5 CMF C4 H12 O4 Si

IT 23432-62-4

RL: PEP (Physical, engineering or chemical process); PROC (Process) (precursor; composition-structure relations in organically modified silica aerogels)

RN 23432-62-4 HCAPLUS

CN Carbamic acid, N-[3-(trimethoxysily1)propy1]-, methyl ester (CA INDEX NAME)

REFERENCE COUNT: 24 THERE ARE 24 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L16 ANSWER 6 OF 7 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1998:505542 HCAPLUS Full-text

DOCUMENT NUMBER: 129:234338

TITLE: Formation and Structure of Porous Gel Networks from

Si(OMe) 4 in the Presence of A(CH2) nSi(OR) 3 (A =

Functional Group)

AUTHOR(S): Huesing, Nicola; Schubert, Ulrich; Misof, Klaus;

Fratzl, Peter

CORPORATE SOURCE: Institut fuer Anorganische Chemie, Technische

Universitaet Wien, Vienna, A-1060, Austria

SOURCE: Chemistry of Materials (1998), 10(10), 3024-3032

CODEN: CMATEX; ISSN: 0897-4756

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal LANGUAGE: English

Monolithic silica aerogels modified by functional org, groups were prepared by sol-gel processing of Si(OMe)4/A(CH2)nSi(OR)3 mixts. under identical exptl. conditions, followed by drying of the wet gels with supercrit. CO2. The employed functional groups A were SH (n = 3), OCH2CHCH2O (n = 3), OC(0)C(Me)=CH2 (n = 3), NCO (n = 3), Cl (n = 3), NHC(0)OMe (n = 3), and PPh2 (n = 2). These groups were retained in the aerogels except the isocyanate groups, which reacted with methanol to the corresponding carbamate. The properties of the obtained aerogels are rather independent of the kind of functional group, but strongly depend on the Si(OMe)4/A(CH2)nSi(OR)3 ratio, which was varied between 9:1 and 6:4. The d. of the aerogels was 0.2-0.3 g.cm-3; some aerogels with lower densities were also prepared for comparison. Gelling of the precursor mixts. is drastically slowed with an increasing portion of A(CH2)nSi(OR)3, and the water consumption is retarded. During supercrit. drying, shrinkage of .apprx.10% was observed for the aerogels prepared from the 9:1 precursor mixts. Increasing the portion of A(CH2)nSi(OR)3 or decreasing the aerogel d. lead to a larger shrinkage and an incomplete incorporation of the functional organic groups. The chemical composition of the resulting aerogels was investigated by IR and Raman

spectroscopy, elemental anal., and titration of the functional groups, and their structure by nitrogen sorption and small-angle X-ray scattering (SAXS). The Brunauer-Emmett-Teller (BET) surface areas generally decreased with an increasing portion of A(CH2)nSi(OR)3, whereas the C parameter showed a saturation behavior. An anal. of the pore vols. indicated that with an increasing portion of A(CH2)nSi(OR)3 or a decreasing bulk d., the gel skeleton is increasingly compressed during the N2 sorption expts. SAXS measurements showed larger particles upon increasing the A(CH2)nSi(OR)3/Si(OMe)4 ratio, which correlates with the observed decrease of the sp. surface areas. The results were interpreted that an increasing portion of A(CH2)nSi(OR)3 has the same kinetic effects on the hydrolysis and condensation reactions and the same structural consequences for the network formation as decreasing the bulk d. of an aerogel obtained from the one-component Si(OMe)4 system. The fractal dimension increased with an increasing portion of A(CH2)nSi(OR)3; it was significantly larger for A = NCO and NHC(O)OMe than for A = SH or OC(O)C(Me):CH2.

IT 23432-62-4

RL: PEP (Physical, engineering or chemical process); PROC (Process) (organic modifier; formation and structure of porous silica gel networks from tetramethoxysilane in the presence of trialkoxysilanes)

RN 23432-62-4 HCAPLUS

CN Carbamic acid, N-[3-(trimethoxysilyl)propyl]-, methyl ester (CA INDEX NAME)

REFERENCE COUNT: 35 THERE ARE 35 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L16 ANSWER 7 OF 7 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1985:167740 HCAPLUS Full-text

DOCUMENT NUMBER: 102:167740

ORIGINAL REFERENCE NO.: 102:26393a,26396a

TITLE: Curable epoxy resin compositions

PATENT ASSIGNEE(S): Shin-Etsu Chemical Industry Co., Ltd., Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 8 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 59197421	A	19841109	JP 1983-73195	19830426
JP 62048968	В	19871016		
PRIORITY APPLN. INFO.:			JP 1983-73195	19830426
GT				

$$\mathbb{R}^{1}\mathbb{N} \longrightarrow \mathbb{N}^{1}$$

AB Moisture-resistant potting compns. for elec. parts comprise a curable epoxy resin 100, an inorg. filler 100-500, and Q(ZR)m, R2(NR1CO)nOR3, or I (Q = H, m-valent organic group; Z = O, NR4; R = CONR1R5; R1 = Z1SiR6aR73-a; R2-R6 = H, monovalent organic group; R7 = OH, hydrolyzable group; Z1 = bivalent organic group; m, n = 1-5; a = 0-2) 0.05-10 parts. Thus, a composition comprising ECN-1280 [63992-68-7] (cresol novolak epoxy resin) 70, MP-120 [9003-35-4] (phenol novolak resin) 30, triphenylphosphine 2, carnauba wax 1, carbon black 1, silica powder 300, and H2NCONH(CH2)3Si(OMe)3 [23843-64-3] 1.5 parts had volume resistivity 2 + 1016 Ω -cm (5 + 1015 Ω -cm after 200 h at 121° and 2.2 atm), flexural strength 15.0 (12.0) kg/cm2, and good stability, adhesion, and processability.

IT 23432-62-4

RL: USES (Uses)

(coupling agents, cresol novolak epoxy resin potting compns. containing)

RN 23432-62-4 HCAPLUS

CN Carbamic acid, N-[3-(trimethoxysily1)propy1]-, methyl ester (CA INDEX NAME)

=> log h		
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